

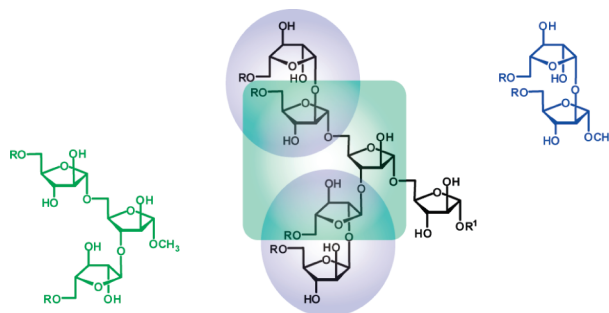
Probing the Effect of Acylation on Arabinofuranose Ring Conformation in Di- and Trisaccharide Fragments of Mycobacterial Arabinogalactan

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Received March 25, 2010



R = H or lipid, R¹ = arabinogalactan

A major component of the cell wall of mycobacteria is the mycolyl–arabinogalactan (mAG) complex. The arabinose and galactose residues in mAG are found solely in the furanose form, and it has been suggested that the flexibility of these five-membered rings allows for the tight packing of mycolic acids. In order to probe the “flexible scaffold hypothesis”, we designed and synthesized glycolipids **3–6** and **8–11** as simple models of the terminal portion of mAG. A set of donors and acceptors were explored for preparing the key β -(1 \rightarrow 2) linkage in **2–6**, and the best selectivity and yield can be obtained by using the electron-rich thioglycoside donor **14** and the O-5 *p*-methoxybenzyl-protected acceptor **17**. Both α -linkages in the trisaccharides **7–11** were formed in a one-pot reaction. The conformations of compounds **2–11** were studied using solution-state NMR spectroscopy, but little change was observed in the coupling constants for the ring protons between **2** and **3–6** or between **7** and **8–11**. However, the rotamer populations about the C-4–C-5 bond for the β -linked ring in disaccharide **2** did change upon acylation at O-5.

Introduction

Tuberculosis remains a major health concern worldwide, with 9.3 million new cases and 1.8 million deaths in 2007.¹ The disease is caused by the bacterium *Mycobacterium tuberculosis*, which produces an unusual cell wall compared to other Gram-positive bacteria.² A second lipid bilayer is

located outside of the plasma membrane,^{3,4} and this bilayer is made up of the unique mycolyl–arabinogalactan (mAG) complex attached to peptidoglycan, as well as various lipids, including trehalose mycolates, phospholipids, and phenolic glycolipids.² The mAG complex is essential for the viability of *M. tuberculosis*;⁵ in fact, two of the front line drugs used to treat tuberculosis inhibit the biosynthesis of mAG.⁶ The mAG is linked to peptidoglycan through a phosphodiester bond to a linker disaccharide of α -D-GlcpNAc-(1 \rightarrow 3)- α -L-Rhap.⁷

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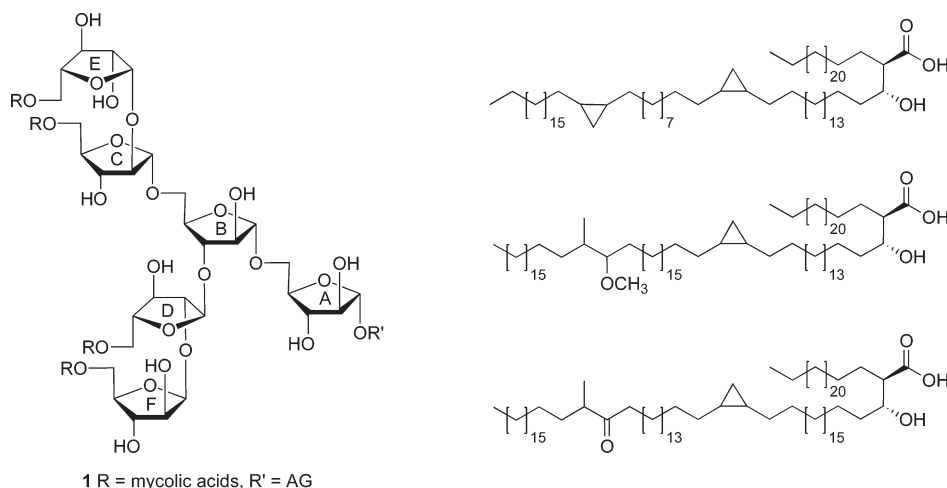


FIGURE 1. (Left) Hexasaccharide motif, **1**, found at the nonreducing end of mycobacterial arabinogalactan (AG). (Right) Three representative examples of mycolic acids found in *M. tuberculosis*.

A linear chain of 30 galactofuranose with alternating β -(1 \rightarrow 5) and β -(1 \rightarrow 6) linkages is attached to this disaccharide, and at residues 8, 10, and 12 are attached arabinan chains.⁸ These arabinose moieties are also in the furanose form, and the 31 monosaccharide residues are linked in a combination of α -(1 \rightarrow 5), α -(1 \rightarrow 3), and β -(1 \rightarrow 2) linkages.^{7,9,10} The arabinan chain is capped at the nonreducing end with the branched hexasaccharide shown in Figure 1 and esterified at O-5 of residues C, D, E, and F with mycolic acids, which are α -alkyl, β -hydroxy lipids with 60–90 carbon atoms.^{11,12}

An interesting feature of the mAG complex is the presence of only furanose sugars, aside from the pyranose residues of the linker disaccharide. Their unusual configuration has led to the proposal that the significant flexibility of the furanose rings facilitates the tight packing of the mycolic acids, the so-called “flexible scaffold hypothesis”.^{13,14} The tight packing of these mycolic acids, in turn, provides a dense, low permeability lipid barrier to the passage of antibiotics into the organism, which facilitates its survival. This is an intriguing hypothesis, but there is little experimental support for it. We reasoned that one approach for testing the flexible scaffold hypothesis would be to compare the conformation of the furanose rings in fragments of mycobacterial arabinan with analogues in which O-5 of the terminal residues had been acylated with lipid chains of varying lengths. It could be expected that, if this hypothesis is correct, conformational differences between the acylated and unacylated compounds would be observed. Thus, we describe here the synthesis of a series of di- and trisaccharide glycolipids (Chart 1) and subsequent conformational analysis of these compounds using

NMR spectroscopy. These relatively simple analogues were chosen as the first targets as this provided us with the opportunity to develop the methodology that could be used to prepare and study more complicated analogues that bear closer resemblance to the natural polymeric glycolipid. Moreover, such investigations were necessary in the interest of carrying out a systematic study, in which a variety of structural features (e.g., the size of the carbohydrate moiety and the length, branching, and functionalization of the lipid portion) were to be explored.

Results and Discussion

The first challenge to be faced was to select a method for the synthesis of the disaccharide targets **2–6**, which contain a 1,2-*cis*-arabinofuranoside (β -arabinofuranoside) linkage. A number of choices exist for the preparation of 1,2-*cis*-arabinofuranosides. For example, previous work^{15,16} from our group has led to the development of an indirect method for preparing β -arabinofuranosides by employing 2,3-anhydrofuranosyl donors, which stereoselectively give the β -glycoside. Another indirect method, the intramolecular aglycon delivery approach, has also been reported to be effective for the preparation of β -arabinofuranosides.^{17–21} A direct method developed by Kim and co-workers²² relies on 2-carboxybenzyl donors. Glycosylations with acceptors having benzoyl groups or hindered secondary alcohols afforded excellent yields of the β -glycosides, whereas benzyl-protected alcohols or primary alcohols gave a mixture of α - and β -arabinofuranosides with no significant stereoselectivity. Recently, donors

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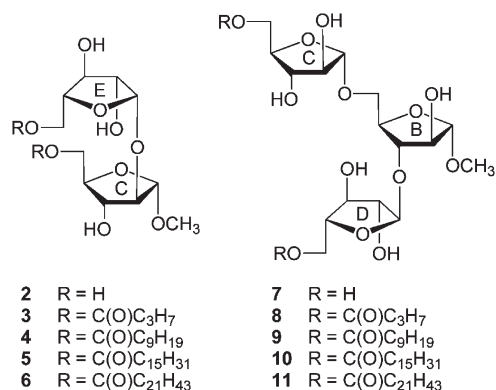
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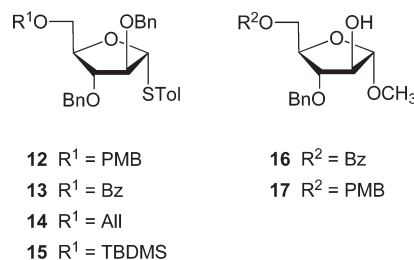
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CHART 1. Target Di- and Trisaccharides (The Rings Have Been Labeled To Facilitate Comparison with 1)

containing a cyclic protecting group have been used to increase the β -stereoselectivity in the preparation of arabinofuranosides. Boons and co-workers^{23–25} and later Crich²⁴ and Zhu^{26,27} have used a di-*tert*-butyl silane (DTBS) acetal to protect the O-3 and O-5 positions in a thioglycoside donor. By rigidifying the flexible furanose ring, the resulting oxocarbenium ion is locked in a conformation that shields the α -face from attack. The β -face, on the other hand, is less hindered; hence, the glycosylation gives excellent β -selectivities with a range of acceptors having either primary or secondary alcohols. In related work, Ito and co-workers²⁸ used a 1,3-(1,1,3,3)-tetraisopropylidisiloxanylidene protecting group to protect the O-3 and O-5 positions of a thioglycoside donor to enhance the β -selectivity.

However, the need to install base-sensitive ester linkages on both of the primary hydroxy groups in the disaccharides made many of these approaches impractical (e.g., the 2,3-anhydrosugar method, or the use of 2-carboxybenzyl donors, which provides high selectivity only when acylated acceptors are used) or cumbersome (e.g., the use of DTBS-protected donors, which would require a series of postglycosylation protection–deprotection steps). Therefore, other approaches were explored.

For the design of the donors, it is desirable to install a group that can be readily removed in the presence of other protecting groups because the O-5 position will be acylated in the final products. In addition, O-2 should be protected with nonparticipating benzyl ether, and to maximize the reactivity of the donor, an electron-rich benzyl group should also be used to protect O-3. With regard to the acceptor, the O-3 position should be protected with benzyl ether as its electron-donating nature will increase the nucleophilicity of hydroxyl group at C-2, and O-5 should be protected with groups that could be removed in the presence of benzyl groups. Keeping in mind these criteria, a number of potential donor and acceptor pairs with different electron-donating/

CHART 2. Donors (12–15) and Acceptors (16 and 17) Used in β -Arabinofuranosylation Reactions

withdrawing protecting groups on O-5 can be envisioned. We describe here the development and optimization of this methodology with various donors (**12–15**) and acceptors (**16** and **17**, Chart 2) and its application to the synthesis of a panel of disaccharide targets. With this panel of compounds, it was possible to probe the effect of the protecting group on O-5 of the donor and acceptor on reaction stereo-selectivity.

Preparation of Thioglycoside Donors 12–15. The thioglycoside donors **12** and **13** were synthesized according to the literature.^{28–30} Donors **14** and **15** were synthesized from known compound **18**^{16,29} (Scheme 1). The free O-5 hydroxyl group in **18** was reacted with sodium hydride and allyl bromide to give donor **14** in excellent (91%) yield. Reaction of the same hydroxyl group with *tert*-butyldimethylsilyl chloride (TBDMSCl) and imidazole gave donor **15** in 75% yield.

Preparation of Acceptors 16 and 17. With the synthesis of the donors finished, we explored the synthesis of acceptors starting from methyl α -D-arabinofuranoside **19** (Scheme 2). By employing Mitsunobu conditions,³¹ the epoxide ring was introduced by reaction of **19** with diisopropyl azodicarboxylate (DIAD) and triphenylphosphine, which gave exclusively the desired product **20** in 90% yield.

We optimized the method for the introduction of the benzyl group at O-3, which involved opening of the epoxide ring with sodium benzyolate. The best results were obtained when **20** was reacted with 2 equiv of sodium benzyolate at high concentration at 100 °C overnight. Under these conditions, the reaction gave diol **21** in 80% yield. The regioselectivity was confirmed by the proton chemical shift of the corresponding acylated derivative of **21**, which has a significant downfield shift of H-2. By controlling the reaction temperature at –10 °C and the amount of benzoyl chloride, selective protection of **21** at the primary position of the diol gave acceptor **16**, which was obtained in 87% yield.

To synthesize acceptor **17**, epoxide **20** was treated with sodium hydride and *p*-methoxybenzyl chloride (PMBCl) to afford the protected epoxide **22** in 80% yield. As before, ring opening of the epoxide in **23** with sodium benzyolate gave the desired compound **17** in 77% yield.

Glycosylation of Donors 12–15 and Acceptors 16 and 17. For determining optimal conditions to use in the glycosylation reactions, we studied the reaction of **13** and **16** (Table 1). We first evaluated the effect of temperature (entries 1–5) using

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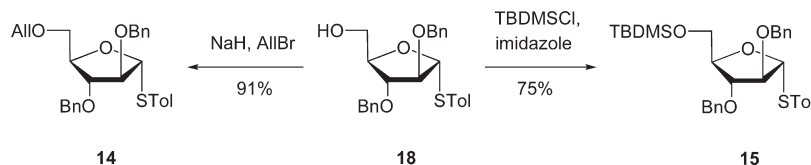
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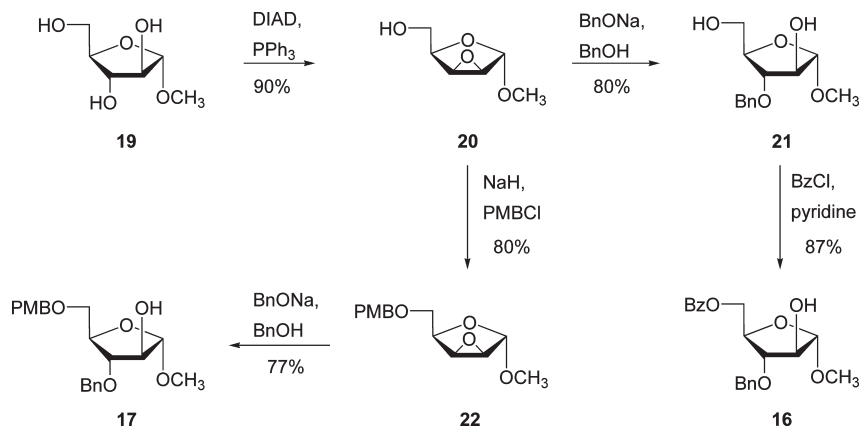
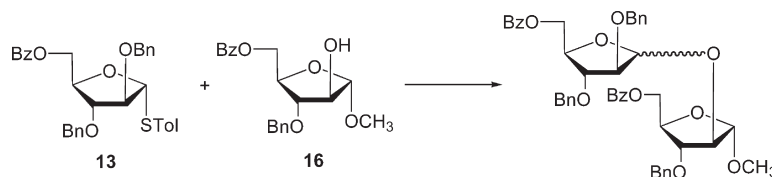
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SCHEME 1. Synthesis of Thioglycoside Donors 14 and 15



SCHEME 2. Synthesis of Acceptors 16 and 17

TABLE 1. Optimization of β -Arabinofuranosylation Using Thioglycoside 13 and Acceptor 16

| entry | solvent | temperature | time (h) | acceptor concentration (M) | activator | yield (α/β) ^d |
|-------|---------------------------------|-----------------|----------|----------------------------|--|---------------------------------------|
| 1 | CH ₂ Cl ₂ | -78 °C → rt | 4 | 0.08 | NIS-AgOTf ^a | 81% (3.1:1) |
| 2 | CH ₂ Cl ₂ | -78 °C | 6 | 0.08 | NIS-AgOTf ^a | 85% (3:1) |
| 3 | CH ₂ Cl ₂ | -60 °C | 4 | 0.08 | NIS-AgOTf ^a | 91% (3.4:1) |
| 4 | CH ₂ Cl ₂ | -40 °C | 0.5 | 0.08 | NIS-AgOTf ^a | 74% (4.6:1) |
| 5 | CH ₂ Cl ₂ | -60 °C → -40 °C | 1 | 0.08 | NIS-AgOTf ^a | 89% (4.2:1) |
| 6 | CH ₂ Cl ₂ | -60 °C → -40 °C | 1 | 1.00 | NIS-AgOTf ^a | 84% (4.3:1) |
| 7 | CH ₂ Cl ₂ | -60 °C → -40 °C | 1.5 | 0.05 | NIS-AgOTf ^a | 85% (4:1) |
| 8 | CH ₂ Cl ₂ | -60 °C → -40 °C | 2 | 0.01 | NIS-AgOTf ^a | 93% (3:1) |
| 9 | CH ₂ Cl ₂ | -60 °C → -40 °C | 0.5 | 0.01 | NIS-TMSOTf ^b | 78% (5:1) |
| 10 | CH ₂ Cl ₂ | -60 °C → -40 °C | 6 | 0.01 | Ph ₂ SO-TTBP-Tf ₂ O ^c | 63% (4:1) |

^aAcceptor **16** (1 equiv), donor **13** (1.2 equiv), NIS (1.2 equiv), AgOTf (0.1 equiv). ^bAcceptor **16** (1 equiv), donor **13** (1.2 equiv), NIS (1.2 equiv), TMSOTf (0.1 equiv). ^cAcceptor **16** (1 equiv), donor **13** (1.2 equiv), Ph₂SO (3 equiv), TTBP (6 equiv), Tf₂O (1.1 equiv). ^dRatio was determined by ¹H NMR spectroscopy.

the standard *N*-iodosuccinimide and silver trifluoromethanesulfonate (NIS-AgOTf) promoter system³² and dichloromethane as the solvent. When the glycosylations were carried out at -78 °C, the reactions were very slow and usually took 5–6 h to finish. Performing the reaction -60 °C led to reduced reaction times (3–4 h). We then found that at -40 °C the reaction proceeded very quickly but that the yield decreased. Thus, we developed an approach where the reaction was started at -60 °C and then slowly warmed to -40 °C over 2 h. Under these conditions, the reaction gave improved yield and stereoselectivity.

The effect of concentration was also studied (entries 6–8) and was shown not to have a dramatic effect on the yield,

although at the lowest concentration investigated (0.01 M), the β -selectivity was somewhat increased. In addition, other glycosylation promoters, such as NIS-trimethylsilyl trifluoromethanesulfonate (TMSOTf)^{33,34} and diphenyl sulfoxide, 2,4,6-tri-*tert*-butylpyrimidine, and trifluoromethanesulfonic anhydride (Ph₂SO-TTBP-Tf₂O),³⁵ were investigated (entries 9 and 10). However, with these promoters, the yield of the product was reduced as was the β -selectivity. Finally, we studied the use of acetonitrile, diethyl ether, toluene, and dioxane as the solvent.

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SCHEME 3. Glycosylation of Donors 12–15 with Acceptors 16 and 17

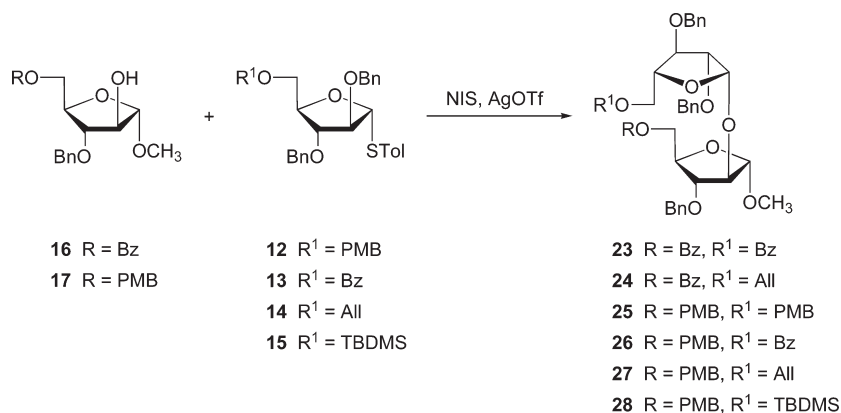


TABLE 2. Glycosylation with Thioglycosides 12–15

| entry | acceptor | donor | product | yield (%) | α/β ratio ^a |
|-------|-----------|-----------|-----------|-----------|-----------------------------------|
| 1 | 16 | 12 | 23 | 96 | 1:3 |
| 2 | 16 | 13 | 24 | 93 | 3:1 |
| 3 | 16 | 14 | 25 | 96 | 1:2.5 |
| 4 | 17 | 12 | 26 | 89 | 1:7.7 |
| 5 | 17 | 13 | 27 | 93 | 1:4.4 |
| 6 | 17 | 14 | 28 | 93 | 1:6.3 |
| 7 | 17 | 15 | 29 | 91 | 1:2 |

^aRatio determined by weight of the isolated pure compounds. ^bThe α/β ratio of products could not be separated; ratio was determined by ¹H NMR spectroscopy.

However, when performed in any of these solvents, the reactions required extended reaction times and gave poor selectivity compared to those carried out in dichloromethane.

The investigations described above allowed us to identify a set of optimized conditions for coupling donors **12**–**15** with acceptors **16** and **17** (Scheme 3). All reactions were carried out at an acceptor concentration of 0.01 M by cooling the donor (1.2 equiv) and acceptor (1 equiv) in dichloromethane at $-60\text{ }^{\circ}\text{C}$, then NIS and AgOTf were added and the reaction was warmed to $-40\text{ }^{\circ}\text{C}$ over 90–120 min. Following workup, the reaction mixture was purified by chromatography, and the isolated anomeric glycosides were characterized. Distinction between α - and β -isomers was achieved by using $\delta_{\text{C-1}}$ and $^3J_{\text{H-1,H-2}}$ values.³⁶ For the β -isomer, $\delta_{\text{C-1}} = 97\text{--}103\text{ ppm}$, $^3J_{\text{H-1,H-2}} = 3\text{--}5\text{ Hz}$, whereas for the α -isomer, $\delta_{\text{C-1}} = 104\text{--}111\text{ ppm}$, $^3J_{\text{H-1,H-2}} = 0\text{--}2\text{ Hz}$. As Table 2 indicates, the yields of these reactions were very good overall, but the α/β ratios changed significantly, depending on the different substituents on the donors and acceptors.

The treatment of the O-5 benzoyl-protected acceptor **16** with any of the donors gave poor to moderate β -selectivity. However, when the *p*-methoxybenzyl-protected acceptor **17** was used, the β/α ratios were enhanced and moderate to good β -selectivity was observed. In other words, the more electron-rich acceptor favored the formation of the β -glycoside. As expected, the structure of the donor also greatly influenced the selectivity. For instance, the ester-protected donor **13** gave poor to moderate selectivity, whereas the two ether-protected species (**12** and **14**) gave better β -selectivity; the *p*-methoxybenzyl ether, **12**, provides slightly more of the β -anomer than the allyl ether, **14**. Finally, the silyl-substituted

donor **15** gave poor β -selectivity. One result that remained unclear was why the more electron-rich, “armed” donors **12** and **14** gave better β -selectivity than the less electron-rich, “disarmed” one, **13**. Armed donors³⁷ are more reactive; thus, the glycosylation should proceed faster and prefer an oxocarbenium ion intermediate resulting in the formation of α/β mixtures. However, in our case, the combination of the most active donor **12** and acceptor **17** gave the best β/α ratio. This finding is consistent with earlier work both from the Ito laboratory^{28,29} as well as from our laboratory.^{38,39}

Synthesis of Disaccharide Targets 2 and 30. Once the optimization of the β -arabinofuranosyl reaction was complete, we then synthesized the target disaccharides. As indicated in Table 2, treatment of acceptor **17** with donor **12** yielded the best α/β ratio, 1:7.7. However, the separation of the isomers was difficult; therefore, the combination of acceptor **17** and donor **14** was chosen to synthesize the disaccharides. Although the reaction was a bit less stereoselective, 1:6.3 α/β , the α - and β -isomers were easily separated after the removal of the allyl group (Scheme 4). With a route to the pure β -isomer **29** in place, reaction with trifluoroacetic acid cleaved the *p*-methoxybenzyl ether affording **30**, which was ready for acylation. Alternatively, hydrogenolysis of **29** with 20% palladium hydroxide on carbon gave the fully deprotected disaccharide **2** in 85% yield.

Preparation of Trisaccharides 7 and 34. In comparison with the synthesis of the β -linked disaccharide, the target trisaccharides had two α -glycosidic linkages, which can be readily constructed by the use of donors with participating acyl groups on O-2. In designing the synthetic approach to the trisaccharides, we chose an efficient and convergent route, in which the target was assembled by the coupling of a diol acceptor and a single donor to add both α -arabinofuranosyl residues in a single reaction.

With this consideration in mind, trisaccharide **7** was synthesized according to the literature.⁴⁰ Similarly, the O-5-deprotected trisaccharide **34** was assembled from known building blocks **31**⁴⁰ and **32**³⁸ in a straightforward way by glycosylation at $0\text{ }^{\circ}\text{C}$ using NIS–AgOTf as the promoter (Scheme 5). Reaction

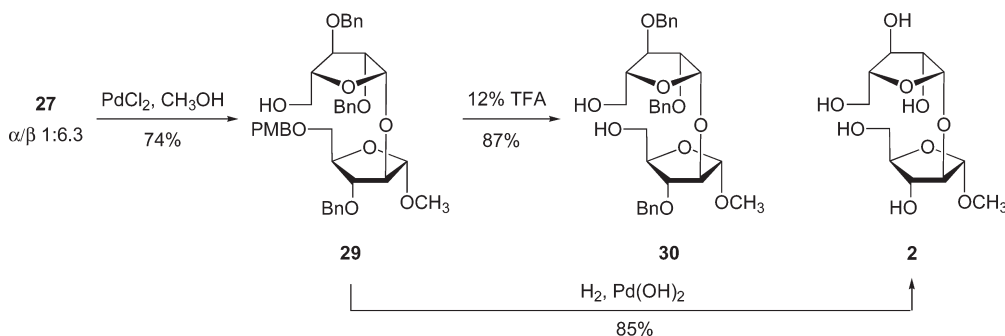
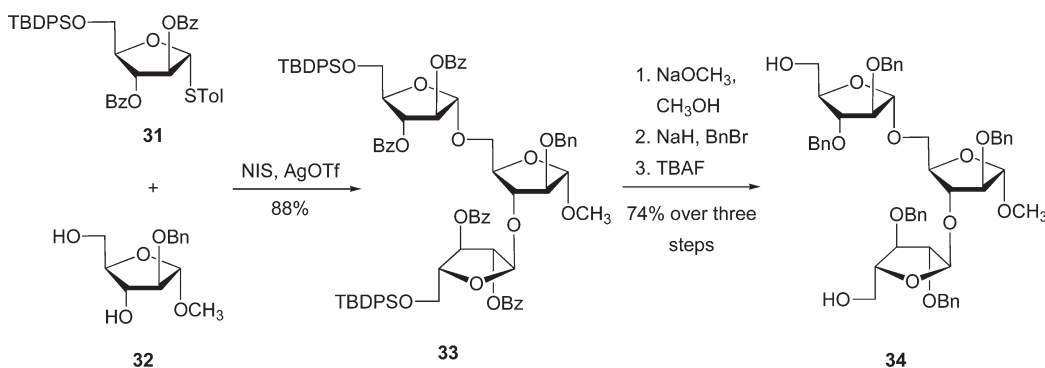
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SCHEME 4. Synthesis of Disaccharides **2** and **30**SCHEME 5. Synthesis of Trisaccharide **34**

of an excess donor **31** with diol **32** afforded the desired trisaccharide **33** in 88% yield. Conversion of **33** to the target diol **34** was carried out by subsequent removal of the benzoyl groups, benzylation, and cleavage of the silyl ethers, in 74% over three steps. Confirmation of the stereochemistry was again possible by using ¹H and ¹³C NMR spectroscopy. For instance, in **34**, the three anomeric signals in the ¹H NMR spectrum, 5.17 ppm (singlet), 5.13 ppm (doublet, *J* = 1.5 Hz), 4.96 ppm (doublet, *J*_{H-1,H-2} = 1.5 Hz), indicated the presence of three α-arabinofuranosyl linkages. In the ¹³C NMR spectrum of **34**, these signals are correlated to resonances at 106.9, 106.2, and 105.9 ppm, respectively, by HMQC, which are also indicative of the α-stereochemistry.

Synthesis of Target Glycolipids. Having the sugar moieties of the glycolipid targets in hand, we next explored their coupling with fatty acid counterparts to obtain the target glycolipids. Three carboxylic acid chlorides and one carboxylic acid were chosen to carry out the reactions: butyryl chloride, decanoyl chloride, palmitoyl chloride, and behenic acid. The choice of acyl chloride versus carboxylic acid was made based on commercial availability. The difference between each member of this homologous series of lipids is six carbon atoms.

As expected, the coupling reactions were carried out without problem through the reaction of the disaccharide or trisaccharide diol with the carboxylic acid chloride in pyridine or with the carboxylic acid in the presence of *N,N'*-dicyclohexylcarbodiimide (DCC). The acylated compounds were next debenzylated using palladium hydroxide on carbon under a hydrogen atmosphere to give the target molecules in 68–93% over the two steps, as shown in Scheme 6.

The solvent system used in the debenzylation reactions was a 1:1 mixture of dichloromethane and methanol. Other solvent systems, such as methanol, a mixture of THF and

methanol, and a mixture of methanol and ethyl acetate, were also tested. However, only the dichloromethane–methanol system gave a good product yield in a reasonable reaction time, usually overnight, at room temperature. Also, to avoid the possible formation of micelles of the partially deprotected compounds, which we anticipated would slow the heterogeneous debenzylation reaction, the concentration of the substrate was kept low, below 0.01 M.

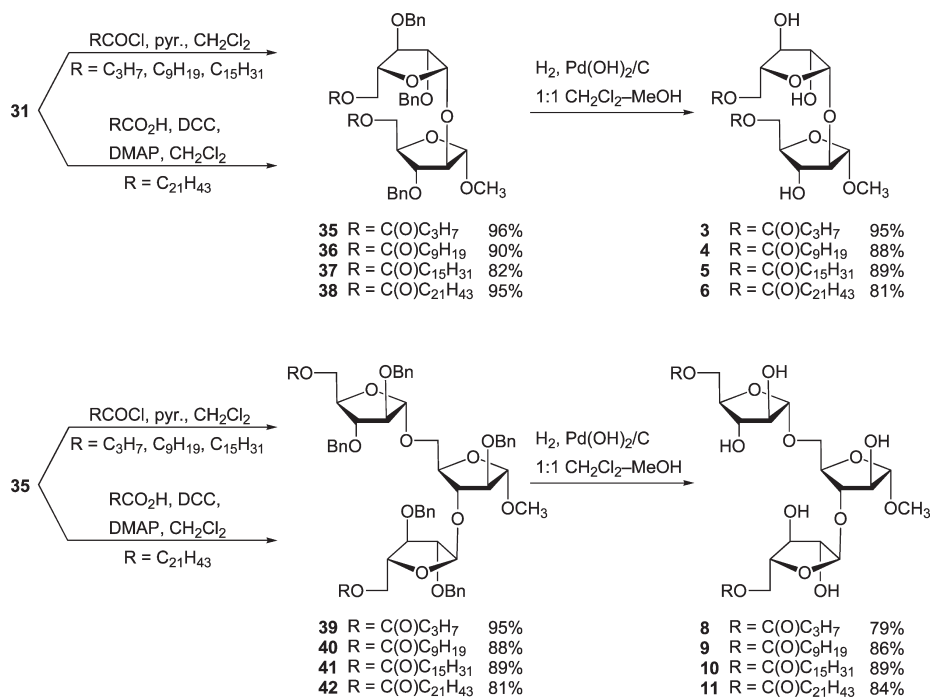
NMR Analysis for the Measurement of Coupling Constants. While compounds **2** and **7** are fully soluble in D₂O, none of the glycolipids **3–6** and **8–11** are, which limited the choice of solvents available for conformational analysis. Thus, in all experiments, NMR spectra were acquired using a solution of the target compound in CD₃OD or 1:1 CDCl₃–CD₃OD. As outlined below, the choice of solvent had negligible effect on conformation and, in turn, coupling constant values. The assignment process is summarized below.

For the assignment of resonances in the spectra of the oligosaccharides, we followed the “structural reporter group” concept developed by Vliegthart and co-workers.⁴¹ The α- and β-arabinofuranosides have distinct and well-resolved signals in this region of the spectrum, and the stereochemistry of glycosidic linkages can be determined from their ³*J*_{H-1,H-2} values, as described above. These proton resonances are correlated with carbon signals (110–90 ppm) by using 2D HMQC or HSQC experiments. The remainder of the ring protons were assigned by means of 2D COSY and TOCSY experiments. In addition, all of the 1D ¹H NMR spectra were simulated using the program WinDNMR⁴² to extract accurate

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SCHEME 6. Synthesis of Disaccharide Glycolipids 3–6 and Trisaccharide Glycolipids 8–11



$^2J_{\text{H,H}}$, $^3J_{\text{H,H}}$, or $^4J_{\text{H,H}}$ values. The coupling constants for the disaccharides are shown in Table 3, and those for the trisaccharides are given in Table 4. The chemical shifts can be found in the Supporting Information, along with representative examples of the simulated spectra. Assignment of H-5R and H-5S was based on comparison with previous literature,^{43,44} and the $^2J_{\text{H-5R,H-5S}}$ values are assumed to be negative.⁴⁵ At the concentrations used to measure the coupling constants, no broadening of signals was observed, suggesting that the compounds are not aggregating.

The data above demonstrate that acylation has no significant effect on ring conformation as the values for $^3J_{\text{H-1,H-2}}$, $^3J_{\text{H-2,H-3}}$, $^3J_{\text{H-3,H-4}}$, and $^4J_{\text{H-1,H-3}}$ do not change between compound **2** and **3–6** or between **7** and **8–11**. Unsurprisingly, when the coupling constants are analyzed in a standard program for determining furanose ring conformation, PSEUROT,⁴⁶ the same ring conformations and populations are predicted for **2** as for **3–6**, as well as for **7** and **8–11** (data not shown). Additionally, changing the NMR solvent from CD_3OD to 1:1 $\text{CD}_3\text{OD}-\text{CDCl}_3$ for **3** and **10** appears to have no significant effect on ring conformation. Because the chemical shifts for the ring protons do not change upon acylation of O-5, we assume that rotation about the glycosidic linkages also remains the same for all of the compounds.

The values for $^3J_{\text{H-4,H-5R}}$ and $^3J_{\text{H-4,H-5S}}$ appear to change more significantly upon acylation, and thus we decided to investigate the rotamer populations about the C-4–C-5 bond in more detail. We focused on rings C, D, and E, which have free 5-OH groups in **2** or **7**; the conformation about the

TABLE 3. $^2J_{\text{H,H}}$, $^3J_{\text{H,H}}$, and $^4J_{\text{H,H}}$ (Hz) Values for Disaccharide **2** and Disaccharide Glycolipids **3–6**

| ring | coupling | 2 ^a | 3 ^a | 3 ^b | 4 ^b | 5 ^b | 6 ^b |
|--------------------------|--------------------------|-----------------------|-----------------------|-----------------------|-----------------------|-----------------------|-----------------------|
| C | $^3J_{\text{H-1,H-2}}$ | 2.2 | 2.1 | 2.1 | 2.1 | 2.1 | 2.1 |
| | $^4J_{\text{H-1,H-3}}$ | 0.7 | 0.0 | 0.0 | 0.0 | 0.5 | 0.0 |
| | $^3J_{\text{H-2,H-3}}$ | 5.0 | 4.6 | 4.6 | 4.6 | 4.7 | 4.6 |
| | $^3J_{\text{H-3,H-4}}$ | 7.6 | 7.5 | 7.5 | 7.5 | 7.5 | 7.4 |
| | $^3J_{\text{H-4,H-5R}}$ | 5.0 | 6.3 | 6.3 | 6.4 | 6.3 | 6.4 |
| | $^3J_{\text{H-4,H-5S}}$ | 2.8 | 3.1 | 3.1 | 3.1 | 3.1 | 3.1 |
| E | $^2J_{\text{H-5R,H-5S}}$ | -12.1 | -12.0 | -11.9 | -12.0 | -12.0 | -11.9 |
| | $^3J_{\text{H-1,H-2}}$ | 4.4 | 4.5 | 4.5 | 4.5 | 4.4 | 4.5 |
| | $^4J_{\text{H-1,H-3}}$ | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 | 0.0 |
| | $^3J_{\text{H-2,H-3}}$ | 8.0 | 7.8 | 7.8 | 7.7 | 7.4 | 7.4 |
| | $^3J_{\text{H-3,H-4}}$ | 6.8 | 7.2 | 7.2 | 7.2 | 7.3 | 7.3 |
| | $^3J_{\text{H-4,H-5R}}$ | 6.2 | 7.2 | 6.3 | 6.3 | 7.5 | 6.7 |
| | $^3J_{\text{H-4,H-5S}}$ | 3.1 | 3.8 | 4.5 | 4.9 | 3.3 | 3.8 |
| $^2J_{\text{H-5R,H-5S}}$ | -12.1 | -11.8 | -12.0 | -12.0 | -11.7 | -11.8 | |

^a CD_3OD as solvent. ^b $\text{CD}_3\text{OD}-\text{CDCl}_3$ (1:1) as solvent.

C-4–C-5 bond of ring B in the trisaccharides is expected to remain unchanged between **7** and **8–11**. Using methods described previously,^{43,44,47} we calculated rotamer populations about the C-4–C-5 bond for the terminal rings in compounds **2–6** and **7–11**. Equations 1 and 2, Karplus relationships⁴⁸ derived specifically for the arabinofuranose ring,⁴⁹ were used for the parent saccharides **2** and **7**.

$$^3J_{\text{H-4,H-5R}} = 5.23 + 0.02\cos(\phi + 15.1^\circ) + 4.67\cos(2\phi + 30.2^\circ) \quad (1)$$

$$^3J_{\text{H-4,H-5S}} = 4.95 - 0.42\cos(\phi) + 4.03\cos(2\phi) \quad (2)$$

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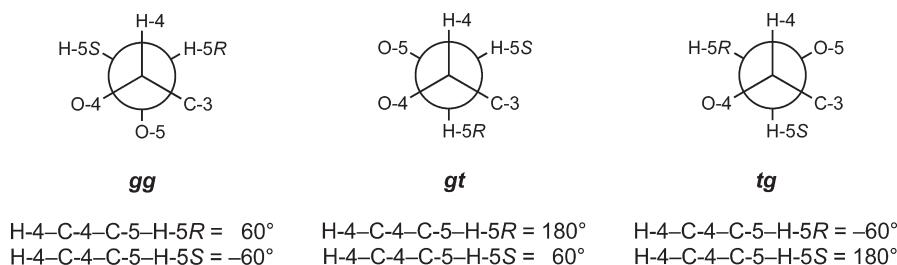


FIGURE 2. Three staggered rotamers about the C-4-C-5 bond of arabinofuranose.

TABLE 4. ${}^2J_{H,H}$, ${}^3J_{H,H}$, and ${}^4J_{H,H}$ (Hz) Values for Trisaccharide 7 and Trisaccharide Glycolipids 8–11

| ring | coupling | 7 ^a | 8 ^a | 9 ^a | 10 ^a | 10 ^b | 11 ^b |
|------|---------------------|----------------|----------------|----------------|-----------------|-----------------|-----------------|
| B | ${}^3J_{H-1,H-2}$ | 1.4 | 1.5 | 1.5 | 1.5 | 1.2 | 1.2 |
| | ${}^4J_{H-1,H-3}$ | 0.7 | 0.6 | 0.5 | 0.6 | 0.6 | 0.6 |
| | ${}^3J_{H-2,H-3}$ | 2.8 | 2.9 | 2.9 | 2.9 | 2.4 | 2.4 |
| | ${}^3J_{H-3,H-4}$ | 6.0 | 6.1 | 6.0 | 6.2 | 5.5 | 5.5 |
| | ${}^3J_{H-4,H-5R}$ | 3.2 | 3.0 | 2.9 | 2.8 | 3.0 | 3.0 |
| | ${}^3J_{H-4,H-5S}$ | 4.9 | 5.1 | 4.9 | 4.9 | 4.4 | 4.5 |
| C | ${}^2J_{H-5R,H-5S}$ | -11.1 | -11.1 | -11.2 | -11.2 | -11.1 | -11.2 |
| | ${}^3J_{H-1,H-2}$ | 1.6 | 1.7 | 1.7 | 1.6 | 1.6 | 1.6 |
| | ${}^4J_{H-1,H-3}$ | 0.7 | 0.6 | 0.7 | 0.6 | 0.7 | 0.5 |
| | ${}^3J_{H-2,H-3}$ | 3.4 | 3.6 | 3.6 | 3.5 | 3.4 | 3.3 |
| | ${}^3J_{H-3,H-4}$ | 6.0 | 6.2 | 6.1 | 6.2 | 5.8 | 5.8 |
| | ${}^3J_{H-4,H-5R}$ | 5.4 | 6.0 | 6.0 | 6.2 | 6.1 | 6.1 |
| D | ${}^3J_{H-4,H-5S}$ | 3.4 | 3.6 | 3.6 | 3.6 | 3.5 | 3.5 |
| | ${}^2J_{H-5R,H-5S}$ | -11.9 | -11.8 | -11.8 | -11.7 | -11.5 | -11.5 |
| | ${}^3J_{H-1,H-2}$ | 1.7 | 1.8 | 1.8 | 1.8 | 1.8 | 1.8 |
| | ${}^4J_{H-1,H-3}$ | 0.6 | 0.7 | 0.7 | 0.5 | 0.7 | 0.6 |
| | ${}^3J_{H-2,H-3}$ | 3.8 | 4.1 | 4.0 | 4.0 | 4.0 | 3.9 |
| | ${}^3J_{H-3,H-4}$ | 6.6 | 7.1 | 7.0 | 6.9 | 6.4 | 6.6 |
| E | ${}^3J_{H-4,H-5R}$ | 5.5 | 6.1 | 6.2 | 6.3 | 6.2 | 6.2 |
| | ${}^3J_{H-4,H-5S}$ | 3.2 | 3.4 | 3.4 | 3.4 | 3.6 | 3.6 |
| | ${}^2J_{H-5R,H-5S}$ | -12.0 | -11.9 | -11.9 | -11.9 | -11.9 | -11.9 |

^aCD₃OD as solvent. ^bCD₃OD-CDCl₃ (1:1) as solvent.

The angle ϕ is the dihedral angle between the coupled protons. These equations were solved to give the maximal coupling constants for the three staggered conformations about the C-4-C-5 bond, where $\phi = 60, 180,$ and -60° (Figure 2).^{40,44,47} The coupling constants measured in solution represent a weighted average of these staggered conformations; the population of each conformation was determined for each ring using eqs 3–5 below:

$${}^3J_{H-4,H-5R} = 1.18X_{gg} + 9.25X_{gt} + 5.26X_{tg} \quad (3)$$

$${}^3J_{H-4,H-5S} = 2.73X_{gg} + 2.73X_{gt} + 9.40X_{tg} \quad (4)$$

$$1 = X_{gg} + X_{gt} + X_{tg} \quad (5)$$

Equations 3–5 were solved for compounds 2 and 7, and the *gg* and *gt* rotamers were found to be favored over the *tg* rotamer, as expected due to the gauche effect.⁵⁰ The results for 2 are in Table 5, and the results for 7 are in Table 6. For the α -linked rings (C of 2, C and D of 7), the *gg* and *gt* rotamers are about equally populated, giving average values of $46 \pm 5\%$ *gg*: $48 \pm 2\%$ *gt*: $6 \pm 5\%$ *tg*. For the β -linked ring E in 2, however, the population of the *gt* rotamer was significantly higher than the *gg* rotamer; the *tg* conformation remained relatively minor: 35% *gg*:59% *gt*:6% *tg*. The C-4-C-5

TABLE 5. Calculated C-4-C-5 Rotamer Distributions (%) for Disaccharides 2–6

| ring | rotamer | 2 ^a | 3 ^a | 3 ^b | 4 ^b | 5 ^b | 6 ^b |
|------|-----------|----------------|----------------|----------------|----------------|----------------|----------------|
| C | <i>gg</i> | 52 | 50 | 50 | 48 | 50 | 48 |
| | <i>gt</i> | 47 | 46 | 46 | 48 | 46 | 48 |
| | <i>tg</i> | 1 | 3 | 3 | 4 | 3 | 4 |
| E | <i>gg</i> | 35 | 21 | 27 | 20 | 24 | 31 |
| | <i>gt</i> | 59 | 57 | 46 | 46 | 61 | 51 |
| | <i>tg</i> | 6 | 21 | 27 | 33 | 15 | 18 |

^aCD₃OD as solvent. ^bCD₃OD-CDCl₃ (1:1) as solvent.

TABLE 6. Calculated C-4-C-5 Rotamer Distributions (%) for Trisaccharides 7–11

| ring | rotamer | 7 ^a | 8 ^a | 9 ^a | 10 ^a | 10 ^b | 11 ^b |
|------|-----------|----------------|----------------|----------------|-----------------|-----------------|-----------------|
| C | <i>gg</i> | 43 | 48 | 48 | 44 | 48 | 48 |
| | <i>gt</i> | 47 | 43 | 43 | 45 | 44 | 44 |
| | <i>tg</i> | 10 | 10 | 10 | 11 | 9 | 9 |
| D | <i>gg</i> | 43 | 49 | 47 | 45 | 44 | 44 |
| | <i>gt</i> | 50 | 44 | 45 | 46 | 45 | 45 |
| | <i>tg</i> | 7 | 7 | 8 | 8 | 11 | 11 |

^aCD₃OD as solvent. ^bCD₃OD-CDCl₃ (1:1) as solvent.

rotamers for the parent disaccharides 2 and 7 are comparable to those for the monosaccharides methyl α -D-arabinofuranoside and methyl β -D-arabinofuranoside. In CD₃OD, we found methyl α -D-arabinofuranoside is 45% *gg*: 47% *gt*: 8% *tg* and methyl β -D-arabinofuranoside is 19% *gg*: 65% *gt*: 16% *tg*. Similar populations have been reported for both monosaccharides in D₂O, as well.^{47,51}

For the acylated saccharides 3–6 and 8–11, we developed new Karplus-type relationships for ${}^3J_{H-4,H-5R}$ and ${}^3J_{H-4,H-5S}$ in arabinofuranose rings using DFT calculations; the methods are detailed in the Supporting Information. The newly derived eqs 6 and 7 describe the ${}^3J_{H-4,H-5R}$ and ${}^3J_{H-4,H-5S}$ relationships for both anomers of methyl 5-O-acetyl- α -D-arabinofuranoside.

$${}^3J_{H-4,H-5R} = 5.27 - 0.51\cos(\phi) + 4.83\cos(2\phi) + 0.02\sin(\phi) - 0.04\sin(2\phi) \quad (6)$$

$${}^3J_{H-4,H-5S} = 5.34 - 0.62\cos(\phi) + 4.50\cos(2\phi) + 0.07\sin(\phi) + 2.06\sin(2\phi) \quad (7)$$

Again, the angle ϕ is the dihedral angle between the coupled protons. These equations were solved to give the coupling constants for the staggered conformations about the

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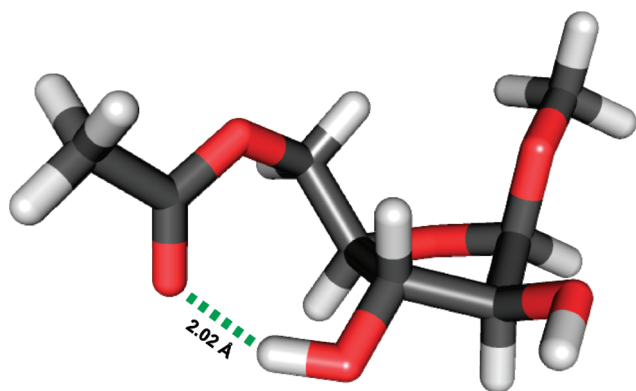


FIGURE 3. Methyl 5-*O*-acetyl- β -D-arabinofuranoside showing the hydrogen bond (green dashed line) from the 3-OH to the carbonyl oxygen of the acetate. This conformation is 2.6 kcal/mol lower in energy than the corresponding conformation without a hydrogen bond, and the distance between the hydrogen and carbonyl oxygen is 2.02 Å. The single point energy was calculated using B3LYP/6-31G(d)^{52,53} in Gaussian03.⁵⁴ The image was created in PyMol.⁵⁵

C-4–C-5 bond, as described above, to give eqs 8–10 below:

$${}^3J_{\text{H-4,H-5R}} = 2.58X_{gg} + 10.62X_{gt} + 2.63X_{tg} \quad (8)$$

$${}^3J_{\text{H-4,H-5S}} = 4.50X_{gg} + 1.05X_{gt} + 10.46X_{tg} \quad (9)$$

$$1 = X_{gg} + X_{gt} + X_{tg} \quad (10)$$

Equations 8–10 were solved for compounds 3–6 and 8–11, and the rotamer populations for the disaccharides 3–6 can be found in Table 5 and those for the trisaccharides 8–11 can be found in Table 6. No significant differences in the rotamer populations were seen for the α -linked ring C of 3–6 or rings C and D of 8–11 when compared to the same rings in 2 and 7. In the acylated species, the *gg* and *gt* conformations were still favored and were present in about equal in population, with an overall average population of $47 \pm 2\%$ *gg*: $45 \pm 2\%$ *gt*: $8 \pm 3\%$ *tg* for these rings. However, unlike the α -linked rings, there was a significant change in the β -linked ring E in compounds 3–6. The *tg* population increased from 6% in the unacylated disaccharide 2 to an average 23% in compounds 3–6, with concomitant decreases in both the *gg* and *gt* conformations, to give an average rotamer distribution of $25 \pm 4\%$ *gg*: $52 \pm 7\%$ *gt*: $23 \pm 7\%$ *tg* for the β -linked rings. One possible explanation for the increase of the *tg* rotamer is that in this conformation the carbonyl oxygen of the acyl group is in the proper orientation to form a weak hydrogen bond to the 3-hydroxyl group, as illustrated in Figure 3. Such interactions would also be anticipated to be important in the relatively hydrophobic outer leaflet of the mycobacterial cell wall, where it is anticipated only limited amounts of water would be present.

Conclusions

In conclusion, we describe here the synthesis of a series of glycolipids (3–6 and 8–11) as simple mimics of the non-

reducing end of the mAG complex present in the mycobacterial cell wall. A key step in the synthesis of the disaccharides was the use of an allyl-protected thioglycoside donor with a *p*-methoxybenzyl-protected acceptor to afford the desired β -(1→2) linkage with good selectivity and high yield. Both the α -(1→3) and α -(1→5) linkages of the trisaccharides were formed in excellent yield using a one-pot protocol.

We analyzed the conformation of the final compounds using solution-state NMR spectroscopy to determine all proton–proton coupling constants, and we determined that there was little effect on the ring conformation upon acylation at O-5. In order to see a change in ring conformation, it may be necessary to use branched acyl chains or explore larger compounds, for example, the hexasaccharide shown in Figure 1, which more closely mimic the natural system. We did, however, see a significant increase in the *tg* rotamer about the C-4–C-5 bond in the β -arabinofuranoside rings present in disaccharides 3–6 upon acylation, suggesting that the lipid chain has some effect on the conformation of the hydroxymethyl group.

Experimental Section

General Procedure for Hydrogenolysis Leading to 2–6 and 8–11. To a solution of starting material (0.02 mmol) in CH_2Cl_2 (1 mL) and CH_3OH (1 mL) was added 20% $\text{Pd}(\text{OH})_2\text{-C}$ (0.15–0.20 fold by weight) at room temperature. The reaction was stirred under a positive pressure of hydrogen for 12 h. The resulting mixture was filtered through Celite and concentrated. The crude residue was purified on Iatrobeads with $\text{CH}_2\text{Cl}_2\text{-CH}_3\text{OH}$ as the eluent.

General Procedures for Glycosylations Leading to 23–28. The acceptor (0.30 mmol) and donor (0.45 mmol) were dried under vacuum in the presence of P_2O_5 for 2 h. To this mixture were added CH_2Cl_2 (30 mL) and 4 Å molecular sieves (0.10 g). The reaction was allowed to stir at room temperature for 30 min and cooled to -60°C , followed by the addition of *N*-iodosuccinimide (0.36 mmol) and silver triflate (0.03 mmol). After the temperature rose to -40°C , the reaction mixture turned dark red/brown. Triethylamine was added to quench the reaction after TLC indicated the completion of reactions. The solution was then diluted with CH_2Cl_2 and filtered through Celite. The filtrate was concentrated to give a crude residue that was purified by chromatography to obtain the product.

General Procedure for Acylation Using Carboxylic Acid Chlorides Leading to 35–37 and 39–41. To a solution of the alcohol (0.06 mmol) in CH_2Cl_2 (2 mL) and pyridine (0.05 mL) was added the carboxylic acid chloride (0.15 mmol) at 0°C . The reaction was kept stirring for 2 h and quenched by the addition of CH_3OH . The resulting solution was concentrated, and the residue was purified by chromatography (4:1 hexane–EtOAc) to obtain the acylated compound.

General Procedure for Acylation Using Behenic Acid Leading to 38 and 42. To a solution of alcohol (0.10 mmol) in CH_2Cl_2 (3 mL) were added behenic acid (131 mg, 0.38 mmol), DCC (79 mg, 0.38 mol), and DMAP (5 mg) at 0°C . The reaction was kept stirring for 4 h and quenched by the addition of CH_3OH . The resulting solution was concentrated, and the residue was purified by chromatography (10:1 hexane–EtOAc) to obtain the acylated product as a white solid.

Methyl β -D-arabinofuranosyl-(1→2)- α -D-arabinofuranoside (2): Isolated as a colorless oil, 85%; R_f 0.37 (7:1 $\text{CH}_2\text{Cl}_2\text{-CH}_3\text{OH}$); $[\alpha]_{\text{D}} -23.2$ (c 1.0, CH_3OH); $^1\text{H NMR}$ (600 MHz, CD_3OD , δ_{H}) 4.95 (d, 1 H, $J_{1\text{E},2\text{E}} = 4.2$ Hz, H-1E), 4.87 (d, 1 H, $J_{1\text{C},2\text{C}} = 2.1$ Hz, H-1C), 4.06–3.95 (m, 4 H, H-2C, H-3C, H-3E, H-2E), 3.88 (ddd, 1 H, $J_{3\text{C},4\text{C}} = 2.6$ Hz, $J_{4\text{C},5\text{Ca}} = 4.9$ Hz, $J_{4\text{C},5\text{Cb}} = 7.4$ Hz, H-4C), 3.81–3.75 (m, 2 H, H-5Ca, H-4E), 3.72 (dd, 1 H, $J_{4\text{E},5\text{Ea}} = 3.2$ Hz, $J_{5\text{Ea},5\text{Eb}} = 11.8$ Hz, H-5Ea), 3.67–3.60 (m, 2 H, H-5Eb,

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$J_{1C,2C} = 1.3$ Hz, H-1C), 4.77 (s, 1 H, H-1B), 4.30–4.24 (m, 2 H, H-5Ca, H-5 Da), 4.18–4.09 (m, 4 H, H-5Cb, H-5Db, H-4C, H-4B), 4.05 (dd, 1 H, $J_{1B,2B} = 1.2$ Hz, $J_{2B,3B} = 2.3$ Hz, H-2B), 4.04–4.00 (m, 3 H, H-2C, H-2D, H-4D), 3.99 (dd, 1 H, $J_{2B,3B} = 2.3$ Hz, $J_{3B,4B} = 5.5$ Hz, H-3B), 3.94 (dd, 1 H, $J_{4B,5Ba} = 4.5$ Hz, $J_{5Ba,5Bb} = 11.1$ Hz, H-5Ba), 3.82–3.78 (m, 2 H, H-3C, H-3D), 3.68 (dd, 1 H, $J_{4B,5Bb} = 3.0$ Hz, $J_{5Ba,5Bb} = 11.1$ Hz, H-5Bb), 3.34 (s, 3 H, OCH₃), 2.36–2.30 (m, 4 H, acyl CH₂ × 2), 1.64–1.54 (m, 4 H, acyl CH₂ × 2), 1.34–1.16 (m, 48 H, acyl CH₂ × 24), 0.88–0.80 (m, 6 H, acyl CH₃ × 2); ¹³C NMR (125 MHz, 1:1 CDCl₃/CD₃OD, δ_C) 175.6 (C=O × 2), 110.6 (C-1B), 109.6 (C-1C), 109.4 (C-1D), 84.8 (C-3B), 83.4 (C-4D), 83.3 (C-4C), 83.2 (C-4B), 82.7 (C-2D), 82.6 (C-2C), 81.7 (C-2B), 79.1 (C-3D, C-3C), 67.8 (C-5B), 65.4 (C-5C), 65.2 (C-5D), 55.9 (OCH₃), 35.3(2) (acyl CH₂), 35.3(0) (acyl CH₂), 33.2 (acyl CH₂), 30.9(4) (acyl CH₂), 30.9(1) (acyl CH₂), 30.8(8) (acyl CH₂), 30.7(7) (acyl CH₂), 30.7(6) (acyl CH₂), 30.6(1) (acyl CH₂), 30.6(0) (acyl CH₂), 30.5(7) (acyl CH₂), 30.4(3) (acyl CH₂), 30.4(2) (acyl CH₂), 26.1 (acyl CH₂), 23.9 (acyl CH₂), 15.0 (acyl CH₃ × 2); ESIMS m/z calcd for [C₄₈H₈₈O₁₅Na]⁺ 927.6015, found 927.6014.

Methyl 5-O-behenoyl- α -D-arabinofuranosyl-(1 \rightarrow 3)-[5-O-behenoyl- α -D-arabinofuranosyl-(1 \rightarrow 5)]- α -D-arabinofuranoside (11): Isolated as a colorless oil, 84%; R_f 0.60 (10:1 CH₂Cl₂–CH₃OH); $[\alpha]_D -3.2$ (c 1.3, CH₃OH); ¹H NMR (600 MHz, 1:1 CDCl₃–CD₃OD, δ_H) 5.00 (d, 1 H, $J_{1D,2D} = 1.7$ Hz, H-1D), 4.97 (d, 1 H, $J_{1C,2C} = 1.2$ Hz, H-1C), 4.77 (s, 1 H, H-1B), 4.30–4.24 (m, 2 H, H-5Ca, H-5 Da), 4.19–4.10 (m, 4 H, H-5Cb, H-5Db, H-4C, H-4B), 4.05 (dd, 1 H, $J_{1B,2B} = 1.2$ Hz, $J_{2B,3B} = 2.4$ Hz, H-2B), 4.04–4.01 (m, 3 H, H-2C, H-2D, H-4D), 3.99 (ddd, 1 H, $J_{1B,3B} = 0.5$ Hz, $J_{2B,3B} = 2.4$ Hz, $J_{3B,4B} = 5.5$ Hz, H-3B), 3.94 (dd, 1 H, $J_{4B,5Ba} = 4.5$ Hz, $J_{5Ba,5Bb} = 11.2$ Hz, H-5Ba), 3.82–3.77 (m, 2 H, H-3C, H-3D), 3.68 (dd, 1 H, $J_{4B,5Bb} = 2.9$ Hz, $J_{5Ba,5Bb} = 11.2$ Hz, H-5Bb), 3.35 (s, 3 H, OCH₃), 2.36–2.30 (m, 4 H, CH₂ × 2), 1.64–1.54 (m, 4 H, CH₂ × 2), 1.40–1.16 (m, 72 H, CH₂ × 36), 0.88–0.80 (m, 6 H, CH₃ × 2); ¹³C NMR (125 MHz, 1:1 CDCl₃–CD₃OD, δ_C) 175.6 (C=O × 2), 110.6 (C-1B), 109.6 (C-1C), 109.4 (C-1D), 84.8 (C-3B), 83.4 (C-4D), 83.3 (C-4C), 83.2 (C-4B), 82.7 (C-2D), 82.6 (C-2C), 81.7 (C-2B), 79.1 (C-3D, C-3C), 67.8 (C-5B), 65.4 (C-5C), 65.2 (C-5D), 55.9 (OCH₃), 35.3(2) (acyl CH₂), 35.3(0) (acyl CH₂), 33.2 (acyl CH₂), 30.9(4) (acyl CH₂), 30.9(0) (acyl CH₂), 30.7(8) (acyl CH₂), 30.7(7) (acyl CH₂), 30.6(1) (acyl CH₂), 30.4(3) (acyl CH₂), 30.4(2) (acyl CH₂), 26.2 (acyl CH₂), 23.9 (acyl CH₂), 15.0 (acyl CH₃ × 2); ESIMS m/z calcd for [C₆₀H₁₁₂O₁₅Na]⁺ 1095.7893, found 1095.7892.

***p*-Tolyl 5-O-allyl-2,3-di-*O*-benzyl-1-thio- α -D-arabinofuranoside (14):** To a solution of compound **18** (2.34 g, 5.36 mmol) in DMF (25 mL) were added NaH (343 mg, 8.58 mmol, 60% dispersion in oil) and AllBr (0.70 mL, 8.04 mmol) at 0 °C. The mixture was stirred at this temperature for 2 h, and the reaction was quenched by the addition of CH₃OH. The solution was diluted with EtOAc and washed with water. The organic layer was dried (MgSO₄), concentrated, and purified by chromatography (8:1 hexane–EtOAc) to afford **14** (2.33 g, 91%) as a colorless syrup: R_f 0.54 (4:1 hexane–EtOAc); $[\alpha]_D +120.4$ (c 1.6, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.42–7.28 (m, 12 H, Ar), 7.12–7.10 (m, 2 H, Ar), 5.93–5.84 (m, 1 H, CH₂=CH–CH₂O), 5.53 (d, 1 H, $J_{1,2} = 3.0$ Hz, H-1), 5.30–5.24 (m, 1 H, CH₂=CH–CH₂O), 5.20–5.16 (m, 1 H, CH₂=CH–CH₂O), 4.68–4.50 (m, 4 H, PhCH₂O), 4.40–4.35 (m, 1 H, H-4), 4.13 (dd, 1 H, $J_{1,2} = J_{2,3} = 3.0$ Hz, H-2), 4.06–3.98 (m, 3 H, CH₂=CH–CH₂O × 2, H-3), 3.67 (dd, 1 H, $J_{4,5a} = 3.9$ Hz, $J_{5a,5b} = 10.9$ Hz, H-5a), 3.61 (dd, 1 H, $J_{4,5b} = 5.1$ Hz, $J_{5a,5b} = 10.9$ Hz, H-5b), 2.34 (s, 3 H, tolyl CH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 137.8 (Ar), 137.4(4) (Ar), 137.3(7) (Ar), 134.6 (Ar), 132.0 (Ar × 2), 131.0 (Ar), 129.7 (CH₂=CH–CH₂O), 128.4(4) (Ar × 2), 128.4(0) (Ar × 2), 128.0 (Ar × 2), 127.9 (Ar × 2), 127.8(4) (Ar), 127.7(9) (Ar × 2), 117.0 (CH₂=CH–CH₂O), 90.6 (C-1), 88.5 (C-2), 83.5 (C-3), 80.4 (C-4), 72.4 (PhCH₂O), 72.3 (PhCH₂O), 72.1 (CH₂=CH–

CH₂O), 69.2 (C-5), 21.1 (tolyl CH₃); ESIMS m/z calcd for [C₂₉H₃₂O₄SiNa]⁺ 499.1914, found 499.1911.

***p*-Tolyl 2,3-di-*O*-benzyl-5-*O*-*tert*-butyldimethylsilyl-1-thio- α -D-arabinofuranoside (15):** To a solution of compound **18** (57 mg, 0.13 mmol) in CH₂Cl₂ (1.3 mL) *tert*-butyldimethylsilyl chloride (50 mg, 0.33 mmol) were added imidazole (45 mg, 0.60 mmol) and DMAP (10 mg). The solution was stirred for 2 h, and then the reaction was quenched by the addition of CH₃OH. The solution was concentrated, and the residue was purified by chromatography (4:1 hexane–EtOAc) to obtain **15** (54 mg, 75%) as a colorless syrup: R_f 0.32 (4:1 hexane–EtOAc); $[\alpha]_D +76.6$ (c 0.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.47–7.41 (m, 2 H, Ar), 7.40–7.29 (m, 10 H, Ar), 7.15–7.11 (m, 2 H, Ar), 5.52 (d, 1 H, $J_{1,2} = 3.0$ Hz, H-1), 4.67 (d, 1 H, $J = 11.9$ Hz, PhCH₂O), 4.60 (s, 2 H, PhCH₂O), 4.54 (d, 1 H, $J = 11.8$ Hz, PhCH₂O), 4.28 (ddd, 1 H, $J_{3,4} = 6.3$ Hz, $J_{4,5a} = 4.4$ Hz, $J_{4,5b} = 4.6$ Hz, H-4), 4.15 (dd, 1 H, $J_{1,2} = J_{2,3} = 3.0$ Hz, H-2), 4.06 (dd, 1 H, $J_{2,3} = 3.0$ Hz, $J_{3,4} = 6.3$ Hz, H-3), 3.82 (dd, 1 H, $J_{4,5a} = 4.4$ Hz, $J_{5a,5b} = 11.0$ Hz, H-5a), 3.79 (dd, 1 H, $J_{4,5b} = 4.6$ Hz, $J_{5a,5b} = 11.0$ Hz, H-5b), 2.35 (s, 3 H, tolyl CH₃), 0.92 (s, 9 H, *t*-butyl CH₃ × 3), 0.08 (s, 6 H, TBDMS CH₃ × 2); ¹³C NMR (125 MHz, CDCl₃, δ_C) 137.9 (Ar), 137.5 (Ar), 137.2 (Ar), 131.9 (Ar × 2), 131.1 (Ar), 129.6 (Ar × 2), 128.4(2) (Ar × 2), 128.3(8) (Ar × 2), 127.9(1) (Ar × 2), 127.8(5) (Ar), 127.7(3) (Ar × 2), 127.7(1) (Ar × 2), 90.5 (C-1), 88.4 (C-2), 83.4 (C-3), 82.2 (C-4), 72.2 (PhCH₂O), 72.1 (PhCH₂O), 62.7 (C-5), 25.9 (*t*-butyl CH₃ × 3), 21.1 (tolyl CH₃), 18.3 (*t*-butyl C), –5.2(7) (Si CH₃), –5.3(4) (Si CH₃); ESIMS m/z calcd for [C₃₂H₄₂O₄SiNa]⁺ 573.2465, found 573.2463.

Methyl 5-*O*-benzoyl-3-*O*-benzyl- α -D-arabinofuranoside (16): Compound **21** (246 mg, 0.97 mmol) was dissolved in CH₂Cl₂ (19 mL) and pyridine (0.47 mL), and the solution was cooled to –10 °C. Benzoyl chloride (0.16 mL, 1.28 mmol) was added dropwise to the mixture over a period of 30 min with stirring. The reaction was subsequently quenched by the addition of CH₃OH, concentrated and the residue purified by chromatography (2:1 hexane–EtOAc) to obtain **16** (301 mg, 87%) as a colorless syrup: R_f 0.32 (2:1 hexane–EtOAc); $[\alpha]_D +90.6$ (c 0.4, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.04–8.01 (m, 2 H, Ar), 7.58–7.54 (m, 1 H, Ar), 7.46–7.41 (m, 2 H, Ar), 7.34–7.24 (m, 5 H, Ar), 4.88 (s, 1 H, H-1), 4.71 (d, 1 H, $J = 12.1$ Hz, PhCH₂O), 4.62 (d, 1 H, $J = 12.1$ Hz, PhCH₂O), 4.55 (dd, 1 H, $J_{4,5a} = 3.1$ Hz, $J_{5a,5b} = 11.5$ Hz, H-5a), 4.43–4.35 (m, 2 H, H-5b, H-4), 4.26 (dd, 1 H, $J_{1,2} = 1.2$ Hz, $J_{2,3} = 2.5$ Hz, H-2), 3.88 (dd, 1 H, $J_{2,3} = 2.5$ Hz, $J_{3,4} = 5.8$ Hz, H-3), 3.42 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.4 (C=O), 137.5 (Ar), 133.1 (Ar), 129.8 (Ar), 129.7 (Ar × 2), 128.5 (Ar × 2), 128.4 (Ar × 2), 128.0 (Ar), 127.9 (Ar × 2), 109.5 (C-1), 85.4 (C-3), 80.6 (C-2), 80.1 (C-4), 72.5 (PhCH₂O), 64.3 (C-5), 55.1 (OCH₃); ESIMS m/z calcd for [C₂₀H₂₂O₆Na]⁺ 381.1309, found 381.1313.

Methyl 3-*O*-benzyl-5-*O*-*p*-methoxybenzyl- α -D-arabinofuranoside (17): To a solution of compound **22** (1.88 g, 7.06 mmol) dissolved in benzyl alcohol (8 mL) was added 1 M BnONa in benzyl alcohol (8.50 mL, 8.50 mmol). The reaction mixture was stirred at 100 °C for 24 h. After cooling to rt, the solution was neutralized with acetic acid, diluted with CH₂Cl₂, and washed with water. The organic layer was then concentrated, and the residue was purified by chromatography (4:1 hexane–EtOAc) to yield **17** (2.03 g, 77%) as a colorless oil: R_f 0.21 (2:1 hexane–EtOAc); $[\alpha]_D +129.0$ (c 0.6, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 7.34–7.27 (m, 5 H, Ar), 7.20–7.16 (m, 2 H, Ar), 6.88–6.84 (m, 2 H, Ar), 4.90 (s, 1 H, H-1), 4.69 (d, 1 H, $J = 12.3$ Hz, PhCH₂O), 4.54 (d, 1 H, $J = 11.4$ Hz, PhCH₂O), 4.52 (d, 1 H, $J = 12.3$ Hz, PhCH₂O), 4.40 (d, 1 H, $J = 11.4$ Hz, PhCH₂O), 4.27 (ddd, 1 H, $J_{3,4} = J_{4,5a} = 2.4$ Hz, $J_{4,5b} = 2.6$ Hz, H-4), 4.11 (d, 1 H, $J_{2,OH} = 10.9$ Hz, H-2), 3.83–3.80 (m, 4 H, H-3, PMBOCH₃), 3.63 (dd, 1 H, $J_{4,5a} = 2.4$ Hz, $J_{5a,5b} = 10.4$ Hz, H-5a), 3.44–3.40 (m, 4 H, H-5b, OCH₃), 3.37 (d, 1 H, $J_{2,OH} = 10.9$ Hz, OH); ¹³C NMR (125 MHz, CDCl₃, δ_C) 159.5 (Ar), 137.7 (Ar), 129.5 (Ar × 2), 129.0 (Ar), 128.4 (Ar × 2), 127.9 (Ar × 2),

127.8 (Ar), 114.0 (Ar × 2), 110.6 (C-1), 85.0 (C-2), 83.7 (C-3), 77.8 (C-4), 73.4 (PhCH₂O), 72.1 (PhCH₂O), 69.4 (C-5), 55.3 (PMBOCH₃, OCH₃); ESIMS *m/z* calcd for [C₂₁H₂₆O₆Na]⁺ 397.1622, found 397.1621.

Methyl 2,3-anhydro-α-D-lyxofuranoside (20). Methyl α-D-arabinofuranoside **19** (2.69 g, 16.4 mmol) and triphenylphosphine (5.16 g, 19.7 mmol) were dissolved in tetrahydrofuran (50 mL), and the solution was cooled to 0 °C. Diisopropylazodicarboxylate (3.88 mL, 19.7 mmol) was added dropwise over a period of 10 min. After complete addition of the reagent, the reaction mixture was warmed to room temperature and was stirred for 1 h. The solution was subsequently concentrated and purified by chromatography (4:1 hexane–EtOAc) to obtain **20** (2.16 g, 90%) as a white crystalline solid: *R_f* 0.14 (2:1 hexane–EtOAc); [α]_D +97.1 (*c* 0.1, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 4.98 (s, 1 H, H-1), 4.17–4.13 (m, 1 H, H-4), 3.91–3.88 (m, 2 H, H-5a, H-5b), 3.75 (dd, 1 H, *J*_{1,2} = 0.7 Hz, *J*_{2,3} = 2.9 Hz, H-2), 3.66 (d, 1 H, *J*_{2,3} = 2.9 Hz, H-3), 3.43 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 102.2 (C-1), 76.2 (C-4), 61.9 (C-5), 55.8 (C-3), 55.6 (OCH₃), 54.1 (C-2); ESIMS *m/z* calcd for [C₆H₁₀O₄Na]⁺ 169.0471, found 169.0472.

Methyl 3-O-benzyl-α-D-arabinofuranoside (21). To a solution of compound **20** (0.72 g, 2.71 mmol) dissolved in benzyl alcohol (4 mL) was added 1 M BnONa in benzyl alcohol (4.10 mL, 4.10 mmol). The reaction mixture was stirred at 100 °C for 24 h. After cooling to rt, the solution was neutralized with acetic acid, diluted with CH₂Cl₂, and washed with water. The organic layer was then concentrated, and the residue was purified by chromatography (4:1 hexane–EtOAc) to yield **21** (0.55 g, 80%) as a colorless syrup: *R_f* 0.38 (2:1 hexane–EtOAc); [α]_D +149.4 (*c* 0.3, CHCl₃); ¹H NMR (600 MHz, CD₃OD, δ_H) 7.35–7.30 (m, 4 H, Ar), 7.26–7.24 (m, 1 H, Ar), 4.79 (s, 1 H, H-1), 4.67 (d, 1 H, *J* = 11.9 Hz, PhCH₂O), 4.55 (d, 1 H, *J* = 11.9 Hz, PhCH₂O), 4.12 (dd, 1 H, *J*_{1,2} = 1.1 Hz, *J*_{2,3} = 2.6 Hz, H-2), 4.03 (ddd, 1 H, *J*_{3,4} = 5.9 Hz, *J*_{4,5a} = 3.6 Hz, *J*_{4,5b} = 5.6 Hz, H-4), 3.76 (ddd, 1 H, *J*_{1,3} = 0.6 Hz, *J*_{2,3} = 2.6 Hz, *J*_{3,4} = 5.9 Hz, H-3), 3.69 (dd, 1 H, *J*_{4,5a} = 3.6 Hz, *J*_{5a,5b} = 11.9 Hz, H-5a), 3.61 (dd, 1 H, *J*_{4,5b} = 5.6 Hz, *J*_{5a,5b} = 11.9 Hz, H-5b), 3.38 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CD₃OD, δ_C) 139.5 (Ar), 129.4 (Ar × 2), 128.9 (Ar × 2), 128.8 (Ar), 111.1 (C-1), 86.8 (C-2), 84.5 (C-3), 81.5 (C-4), 73.1 (PhCH₂O), 63.3 (C-5), 55.1 (OCH₃); ESIMS *m/z* calcd for [C₁₃H₁₈O₅Na]⁺ 277.1047, found 277.1046.

Methyl 2,3-anhydro-5-O-p-methoxybenzyl-α-D-lyxofuranoside (22). To a solution of compound **20** (1.29 g, 8.86 mmol) in DMF (15 mL) were added NaH (425 mg, 10.6 mmol, 60% dispersion in oil) and PMBCl (1.48 mL, 10.6 mmol) at 0 °C. The mixture was stirred at this temperature for 1 h, and the reaction was then quenched by the addition of CH₃OH. The solution was diluted with EtOAc and washed with water. The organic layer was dried (MgSO₄) and concentrated, and the residue was purified by chromatography (6:1 hexane–EtOAc) to afford **22** (1.88 g, 80%) as a colorless oil: *R_f* 0.25 (6:1 hexane–EtOAc); [α]_D +21.8 (*c* 0.7, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.30–7.26 (m, 2 H, Ar), 6.91–6.87 (m, 2 H, Ar), 4.95 (s, 1 H, H-1), 4.55 (d, 1 H, *J* = 11.7 Hz, PhCH₂O), 4.51 (d, 1 H, *J* = 11.7 Hz, PhCH₂O), 4.18 (dd, 1 H, *J*_{4,5a} = *J*_{4,5b} = 6.3 Hz, H-4), 3.81 (s, 3 H, PMBOCH₃), 3.75 (d, 1 H, *J*_{2,3} = 2.3 Hz, H-2), 3.65–3.61 (m, 2 H, H-3, H-5), 3.42 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 159.3 (Ar), 130.0 (Ar), 129.4 (Ar × 2), 113.8 (Ar × 2), 102.3 (C-1), 76.8 (C-4), 73.3 (PhCH₂O), 68.2 (C-5), 56.2 (C-3), 55.6 (OCH₃), 55.3 (PMBOCH₃), 54.4 (C-2); ESIMS *m/z* calcd for [C₁₄H₁₈O₅Na]⁺ 289.1047, found 289.1048.

Methyl 5-O-benzoyl-2,3-di-O-benzyl-β-D-arabinofuranosyl-(1→2)-5-O-benzoyl-3-O-benzyl-α-D-arabinofuranoside (23): Isolated as a colorless oil, 23%; *R_f* 0.35 (3:1 hexane–EtOAc); [α]_D +4.0 (*c* 1.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 8.05–8.02 (m, 2 H, Ar), 7.98–7.95 (m, 2 H, Ar), 7.54–7.50 (m, 2 H, Ar), 7.40–7.18 (m, 19 H, Ar), 5.11 (d, 1 H, *J*_{1E,2E} = 4.5 Hz, H-1E), 4.90 (s, 1 H, H-1C), 4.75–4.68 (m, 2 H, PhCH₂O), 4.64–4.58 (m,

3 H, PhCH₂O), 4.50–4.44 (m, 3 H, PhCH₂O, H-5Ca, H-5Ea), 4.40–4.32 (m, 4 H, H-2C, H-4C, H-5Cb, H-5Eb), 4.27 (ddd, 1 H, *J*_{3E,4E} = *J*_{4E,5Eb} = 6.6 Hz, *J*_{4E,5Ea} = 4.5 Hz, H-4E), 4.25 (dd, 1 H, *J*_{2E,3E} = *J*_{3E,4E} = 6.6 Hz, H-3E), 4.14 (dd, 1 H, *J*_{1E,2E} = 4.5 Hz, *J*_{2E,3E} = 6.6 Hz, H-2E), 4.08 (dd, 1 H, *J*_{2C,3C} = 2.5 Hz, *J*_{3C,4C} = 5.8 Hz, H-3C), 3.40 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.3 (C=O), 166.1 (C=O), 137.7 (Ar), 137.6 (Ar), 137.5 (Ar), 133.7 (Ar), 133.1 (Ar), 133.0 (Ar), 130.2 (Ar), 129.9 (Ar × 2), 129.8 (Ar × 2), 129.7 (Ar × 2), 128.5(5) (Ar × 2), 128.4(7) (Ar), 128.4(4) (Ar), 128.3(5) (Ar × 2), 128.3(4) (Ar × 2), 128.3(0) (Ar × 2), 128.1(3) (Ar × 2), 128.0(7) (Ar), 127.8 (Ar), 127.7(1) (Ar × 2), 127.6(9) (Ar), 107.1 (C-1C), 100.9 (C-1E), 86.3 (C-2C), 84.5 (C-3C), 83.9 (C-2E), 82.3 (C-3E), 80.2 (C-4C), 78.8 (C-4E), 72.8 (PhCH₂O), 72.6(2) (PhCH₂O), 72.5(7) (PhCH₂O), 66.3 (C-5E), 64.4 (C-5C), 54.9 (OCH₃); ESIMS *m/z* calcd for [C₄₆H₄₆O₁₁Na]⁺ 797.2932, found 797.2935.

Methyl 5-O-allyl-2,3-di-O-benzyl-β-D-arabinofuranosyl-(1→2)-5-O-benzoyl-3-O-benzyl-α-D-arabinofuranoside (24): Isolated as a colorless oil, 69%; *R_f* 0.60 (2:1 hexane–EtOAc); [α]_D –5.4 (*c* 0.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.07–8.04 (m, 2 H, Ar), 7.56–7.51 (m, 1 H, Ar), 7.42–7.38 (m, 2 H, Ar), 7.37–7.24 (m, 15 H, Ar), 5.82–5.74 (m, 1 H, CH₂=CH–CH₂O), 5.18–5.13 (m, 1 H, CH₂=CH–CH₂O), 5.10–5.04 (m, 2 H, CH₂=CH–CH₂O, H-1E), 4.99 (s, 1 H, H-1C), 4.76–4.55 (m, 6 H, PhCH₂O), 4.49 (dd, 1 H, *J*_{4C,5Ca} = 3.9 Hz, *J*_{5Ca,5Cb} = 11.7 Hz, H-5Ca), 4.42 (dd, 1 H, *J*_{4C,5Cb} = 5.7 Hz, *J*_{5Ca,5Cb} = 11.7 Hz, H-5Cb), 4.37 (ddd, 1 H, *J*_{3C,4C} = *J*_{4C,5Cb} = 5.7 Hz, *J*_{4C,5Ca} = 3.9 Hz, H-4C), 4.31 (dd, 1 H, *J*_{1C,2C} = 1.1 Hz, *J*_{2C,3C} = 2.6 Hz, H-2C), 4.14–4.06 (m, 4 H, H-2E, H-3E, H-3C, H-4E), 3.91–3.82 (m, 2 H, CH₂=CH–CH₂O), 3.53–3.44 (m, 2 H, H-5Ea, H-5Eb), 3.38 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.3 (C=O), 138.2 (Ar), 137.7 (Ar), 137.6 (Ar), 134.4 (Ar), 133.0 (Ar), 129.9 (Ar), 129.8 (Ar × 2), 129.5 (Ar × 2), 129.3(7) (Ar × 3), 128.3(6) (Ar × 3), 128.1 (Ar), 128.0 (CH₂=CH–CH₂O), 127.7(6) (Ar × 3), 128.7(5) (Ar × 3), 127.7(0) (Ar), 117.1 (CH₂=CH–CH₂O), 107.2 (C-1C), 100.5 (C-1E), 86.0 (C-2C), 84.2 (C-3C), 84.0 (C-2E), 82.8 (C-4C), 80.1 (C-3E), 80.0 (C-4E), 72.6 (PhCH₂O), 72.4(4) (PhCH₂O), 72.4(2) (PhCH₂O), 72.1 (CH₂=CH–CH₂O), 71.9 (C-5E), 64.5 (C-5C), 54.9 (OCH₃); ESIMS *m/z* calcd for [C₄₂H₄₆O₁₀Na]⁺ 733.2983, found 733.2986.

Methyl 2,3-di-O-benzyl-5-O-p-methoxybenzyl-β-D-arabinofuranosyl-(1→2)-3-O-benzyl-5-O-p-methoxybenzyl-α-D-arabinofuranoside (25): Isolated as a colorless oil, 79%; *R_f* 0.28 (3:1 hexane–EtOAc); [α]_D –22.7 (*c* 0.5, CHCl₃); NMR (600 MHz, CDCl₃, δ_H) 7.36–7.01 (m, 17 H, Ar), 7.16–7.12 (m, 2 H, Ar), 6.87–6.84 (m, 2 H, Ar), 6.82–6.78 (m, 2 H, Ar), 5.07 (d, 1 H, *J*_{1E,2E} = 4.2 Hz, H-1E), 4.90 (s, 1 H, H-1C), 4.68–4.64 (m, 2 H, PhCH₂O), 4.62–4.55 (m, 3 H, PhCH₂O), 4.48–4.44 (m, 3 H, PhCH₂O), 4.38–4.30 (m, 2 H, PhCH₂O), 4.26 (dd, 1 H, *J*_{1C,2C} = 1.3 Hz, *J*_{2C,3C} = 2.9 Hz, H-2C), 4.23 (ddd, 1 H, *J*_{3C,4C} = *J*_{4C,5Cb} = 6.0 Hz, *J*_{4C,5Ca} = 4.3 Hz, H-4C), 4.12–4.06 (m, 3 H, H-2E, H-3E, H-4E), 3.99 (dd, 1 H, *J*_{2C,3C} = 2.9 Hz, *J*_{3C,4C} = 6.0 Hz, H-3C), 3.78 (s, 3 H, PMBOCH₃), 3.76 (s, 3 H, PMBOCH₃), 3.58–3.48 (m, 4 H, H-5Ca, H-5Cb, H-5Ea, H-5Eb), 3.38 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 159.1(9) (Ar), 159.1(6) (Ar), 138.2 (Ar), 138.1 (Ar), 137.7 (Ar), 130.3 (Ar), 130.1 (Ar), 129.4 (Ar × 2), 129.2 (Ar × 2), 128.5 (Ar × 2), 128.3(2) (Ar × 2), 129.2(6) (Ar × 2), 127.9(7) (Ar × 2), 127.9(0) (Ar), 127.7(4) (Ar × 2), 127.7(0) (Ar × 2), 127.6(2) (Ar), 127.5(6) (Ar), 113.7 (Ar × 4), 107.0 (C-1C), 100.3 (C-1E), 86.0 (C-2C), 84.0 (C-2E, C-3C), 83.1 (C-3E), 81.4 (C-4C), 80.1 (C-4E), 73.0 (PhCH₂O), 72.8 (PhCH₂O), 72.5 (PhCH₂O), 72.4 (PhCH₂O), 72.2 (PhCH₂O), 71.8 (C-5E), 69.8 (C-5C), 55.2(3) (PMBOCH₃), 55.2(2) (PMBOCH₃), 54.9 (OCH₃); ESIMS *m/z* calcd for [C₄₈H₅₄O₁₁Na]⁺ 829.3558, found 829.3558.

Methyl 5-O-benzoyl-2,3-di-O-benzyl-β-D-arabinofuranosyl-(1→2)-3-O-benzyl-5-O-p-methoxybenzyl-α-D-arabinofuranoside (26): Isolated as a colorless oil, 76%; *R_f* 0.40 (3:1 hexane–EtOAc); [α]_D –11.6 (*c* 2.0, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H)

8.02–7.98 (m, 2 H, Ar), 7.56–7.52 (m, 2 H, Ar), 7.40–7.26 (m, 12 H, Ar), 7.24–7.20 (m, 6 H, Ar), 6.84–6.82 (m, 2 H, Ar), 5.14 (d, 1 H, $J_{1E,2E} = 4.4$ Hz, H-1E), 4.93 (s, 1 H, H-1C), 4.73 (d, 1 H, $J = 11.6$ Hz, PhCH₂O), 4.67–4.55 (m, 4 H, PhCH₂O), 4.48–4.42 (m, 4 H, PhCH₂O, H-5Eb), 4.33 (dd, 1 H, $J_{4E,5Ea} = 6.9$ Hz, $J_{5Ea,5Eb} = 11.4$ Hz, H-5Ea), 4.28–4.24 (m, 2 H, H-2C, H-4E), 4.23–4.17 (m, 2 H, H-4C, H-3E), 4.12 (dd, 1 H, $J_{1E,2E} = 4.4$ Hz, $J_{2E,3E} = 6.9$ Hz, H-2E), 3.98 (dd, 1 H, $J_{2C,3C} = 2.6$ Hz, $J_{3C,4C} = 6.0$ Hz, H-3C), 3.77 (s, 3 H, PMBOCH₃), 3.56–3.50 (m, 2 H, H-5Ca, H-5Cb), 3.39 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.1 (C=O), 159.2 (Ar), 137.9 (Ar), 137.8 (Ar), 137.6 (Ar), 133.1 (Ar × 2), 130.2 (Ar), 129.8 (Ar), 129.7 (Ar × 2), 129.4 (Ar × 2), 128.5 (Ar × 2), 128.4(1) (Ar × 2), 128.3(7) (Ar × 2), 128.2 (Ar × 2), 128.0 (Ar × 3), 127.8 (Ar), 127.7 (Ar × 3), 127.5 (Ar), 113.7(3) (Ar), 113.7(0) (Ar), 107.0 (C-1C), 100.9 (C-1E), 86.8 (C-2C), 84.3 (C-3C), 83.9 (C-2E), 82.5 (C-3E), 81.6 (C-4C), 78.8 (C-4E), 72.9 (PhCH₂O), 72.6 (PhCH₂O × 2), 72.4 (PhCH₂O), 69.8 (C-5C), 66.3 (C-5E), 55.2 (PMBOCH₃), 54.9 (OCH₃); ESIMS m/z calcd for [C₄₇H₅₀O₁₁Na]⁺ 813.3245, found 813.3237.

Methyl 5-O-allyl-2,3-di-O-benzyl- β -D-arabinofuranosyl-(1 \rightarrow 2)-3-O-benzyl-5-O-*p*-methoxybenzyl- α -D-arabinofuranoside (27): Isolated as a colorless oil, 80%; R_f 0.38 (3:1 hexane–EtOAc); $[\alpha]_D -22.7$ (c 0.8, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 7.36–7.23 (m, 17 H, Ar), 6.87–6.84 (m, 2 H, Ar), 5.84–5.76 (m, 1 H, CH₂=CH–CH₂O), 5.20–5.15 (m, 1 H, CH₂=CH–CH₂O), 5.11–5.08 (m, 1 H, CH₂=CH–CH₂O), 5.05 (d, 1 H, $J_{1E,2E} = 3.7$ Hz, H-1E), 4.89 (s, 1 H, H-1C), 4.70–4.55 (m, 5 H, PhCH₂O), 4.51–4.47 (m, 3 H, PhCH₂O), 4.26 (dd, 1 H, $J_{1C,2C} = 1.2$ Hz, $J_{2C,3C} = 2.9$ Hz, H-2C), 4.20 (ddd, 1 H, $J_{3C,4C} = 6.0$ Hz, $J_{4C,5Ca} = J_{4C,5Cb} = 4.2$ Hz, H-4C), 4.10–4.05 (m, 3 H, H-2E, H-3E, H-4E), 3.98 (dd, 1 H, $J_{2C,3C} = 2.9$ Hz, $J_{3C,4C} = 6.0$ Hz, H-3C), 3.91–3.82 (m, 2 H, CH₂=CH–CH₂O), 3.79 (s, 3 H, PMBOCH₃), 3.59–3.53 (m, 2 H, H-5Ca, H-5Ea), 3.51–3.45 (m, 2 H, H-5Cb, H-5Eb), 3.38 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 159.2 (Ar), 138.2 (Ar), 138.1 (Ar), 137.7 (Ar), 134.6 (Ar), 130.2 (Ar), 129.4 (Ar × 3), 128.5 (Ar × 2), 128.4 (Ar × 2), 128.3 (Ar × 2), 128.0 (CH₂=CH–CH₂O), 127.9 (Ar), 127.7(5) (Ar × 4), 127.6(8) (Ar), 127.5(9) (Ar), 116.9 (CH₂=CH–CH₂O), 113.7 (Ar × 2), 107.0 (C-1C), 100.3 (C-1E), 86.0 (C-2C), 84.0(4) (C-2E), 84.0(2) (C-3C), 83.1 (C-3E), 81.3 (C-4C), 80.1 (C-4E), 73.0 (PhCH₂O), 72.5 (PhCH₂O), 72.4 (PhCH₂O), 72.2 (PhCH₂O), 72.1(2) (CH₂=CH–CH₂O), 72.0(8) (C-5E), 69.8 (C-5C), 55.3 (PMBOCH₃), 54.9 (OCH₃); ESIMS m/z calcd for [C₄₃H₅₀O₁₀Na]⁺ 749.3296, found 749.3292.

Methyl 2,3-di-O-benzyl-5-O-*tert*-butyldimethylsilyl- β -D-arabinofuranosyl-(1 \rightarrow 2)-3-O-benzyl-5-O-*p*-methoxybenzyl- α -D-arabinofuranoside (28): Isolated as a colorless oil, 61%; R_f 0.33 (4:1 hexane–EtOAc); $[\alpha]_D -24.6$ (c 0.5, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.38–7.24 (m, 17 H, Ar), 6.89–6.86 (m, 2 H, Ar), 5.08 (d, 1 H, $J_{1E,2E} = 3.8$ Hz, H-1E), 4.93 (s, 1 H, H-1C), 4.73–4.55 (m, 5 H, PhCH₂O), 4.52–4.48 (m, 3 H, PhCH₂O), 4.28–4.25 (m, 1 H, H-2C), 4.24–4.20 (m, 1 H, H-4C), 4.12–4.07 (m, 2 H, H-4E, H-2E), 4.02–3.96 (m, 2 H, H-3E, H-3C), 3.80 (s, 3 H, PMBOCH₃), 3.76–3.66 (m, 2 H, H-5Ea, H-5Eb), 3.60–3.54 (m, 2 H, H-5Ca, H-5Cb), 3.40 (s, 3 H, OCH₃), 0.90–0.87 (m, 9 H, *t*-butyl CH₃ × 3), 0.05–0.02 (m, 6 H, TBDMS CH₃ × 2); ¹³C NMR (125 MHz, CDCl₃, δ_C) 159.2 (Ar), 138.3 (Ar), 138.0 (Ar), 137.8 (Ar), 130.2 (Ar × 2), 129.4 (Ar × 2), 128.4(4) (Ar × 2), 128.3(5) (Ar × 2), 128.2(9) (Ar × 2), 128.0 (Ar × 2), 127.8(7) (Ar), 127.7 (Ar × 2), 127.6(7) (Ar × 2), 127.6(2) (Ar), 113.7 (Ar × 2), 107.0 (C-1C), 100.5 (C-1E), 86.2 (C-2C), 84.1(8) (C-3C), 84.1(6) (C-2E), 84.1(1) (C-4E), 82.2 (C-3E), 81.3 (C-4C), 73.0 (PhCH₂O), 72.5 (PhCH₂O), 72.3(3) (PhCH₂O), 72.3(0) (PhCH₂O), 69.8 (C-5C), 65.4 (C-5E), 55.3 (PMBOCH₃), 54.9 (OCH₃), 25.9 (*t*-butyl CH₃ × 3), 18.3 (*t*-butyl C), –5.2(5) (Si CH₃), –5.3(0) (Si CH₃); ESIMS m/z calcd for [C₄₆H₆₀O₁₀SiNa]⁺ 823.3848, found 823.3852.

Methyl 2,3-di-O-benzyl- β -D-arabinofuranosyl-(1 \rightarrow 2)-3-O-benzyl-5-O-*p*-methoxybenzyl- α -D-arabinofuranoside (29): To a solution

of the disaccharide mixture of **27** (1.45 g, 2.01 mmol) in CH₃OH (50 mL) and CH₂Cl₂ (6 mL) was added PdCl₂ (71 mg, 0.40 mmol). The solution was stirred for 12 h, the reaction was quenched by the addition of triethylamine, and the mixture was concentrated. Chromatography of the residue (8:1 hexane–EtOAc) gave **29** (1.02 g, 74%) as a colorless syrup; R_f 0.32 (2:1 hexane–EtOAc); $[\alpha]_D -23.5$ (c 0.4, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.38–7.24 (m, 17 H, Ar), 6.89–6.86 (m, 2 H, Ar), 5.05 (d, 1 H, $J_{1E,2E} = 4.5$ Hz, H-1E), 4.90 (d, 1 H, $J_{1C,2C} = 1.6$ Hz, H-1C), 4.74 (d, 1 H, $J = 11.6$ Hz, PhCH₂O), 4.66–4.56 (m, 4 H, PhCH₂O), 4.52–4.44 (m, 3 H, PhCH₂O), 4.28 (dd, 1 H, $J_{2E,3E} = 7.1$ Hz, $J_{3E,4E} = 6.5$ Hz, H-3E), 4.23 (dd, 1 H, $J_{1C,2C} = 1.6$ Hz, $J_{2C,3C} = 3.5$ Hz, H-2C), 4.18 (ddd, 1 H, $J_{3C,4C} = 6.3$ Hz, $J_{4C,5Ca} = 4.0$ Hz, $J_{4C,5Cb} = 5.0$ Hz, H-4C), 4.12–4.07 (m, 2 H, H-2E, H-3C), 3.99 (ddd, 1 H, $J_{3E,4E} = 6.4$ Hz, $J_{4E,5Ea} = 3.0$ Hz, $J_{4E,5Eb} = 4.8$ Hz, H-4E), 3.80 (s, 3 H, PMBOCH₃), 3.66–3.50 (m, 4 H, H-5Ea, H-5Ca, H-5Eb, H-5Cb), 3.38 (s, 3 H, OCH₃), 2.29 (dd, 1 H, $J_{5Ea,OH} = 5.0$ Hz, $J_{5Eb,OH} = 7.9$ Hz, OH); ¹³C NMR (125 MHz, CDCl₃, δ_C) 159.3 (Ar), 138.7 (Ar), 137.8 (Ar), 137.6 (Ar), 130.0 (Ar), 129.5 (Ar × 2), 128.5 (Ar × 2), 128.4 (Ar × 2), 128.3 (Ar × 2), 128.0 (Ar × 2), 127.7(9) (Ar × 2), 127.7(8) (Ar × 2), 127.7(3) (Ar), 127.7(0) (Ar × 2), 113.8 (Ar × 2), 107.0 (C-1C), 100.4 (C-1E), 86.6 (C-2C), 84.2 (C-2E), 83.2 (C-3C), 82.0 (C-4E), 81.0 (C-4C), 80.6 (C-3E), 73.0 (PhCH₂O), 72.7 (PhCH₂O), 72.6 (PhCH₂O), 72.2 (PhCH₂O), 69.3 (C-5C), 63.4 (C-5E), 55.3 (PMBOCH₃), 55.1 (OCH₃); ESIMS m/z calcd for [C₄₀H₄₆O₁₀Na]⁺ 709.2983, found 709.2982.

Methyl 2,3-di-O-benzyl- β -D-arabinofuranosyl-(1 \rightarrow 2)-3-O-benzyl- α -D-arabinofuranoside (30): To a solution of compound **29** (0.85 g, 1.24 mmol) in CH₂Cl₂ (6 mL) was added trifluoroacetic acid (0.82 mL). The reaction mixture was allowed to stir for 10 min, and the reaction was then quenched by the addition of triethylamine and the solution concentrated. The residue was purified by chromatography to give **30** (0.61 g, 87%) as a colorless syrup; R_f 0.57 (1:1 hexane–EtOAc); $[\alpha]_D -0.5$ (c 0.4, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 7.39–7.28 (m, 15 H, Ar), 5.00 (d, 1 H, $J_{1E,2E} = 4.6$ Hz, H-1E), 4.87 (s, 1 H, H-1C), 4.74–4.68 (m, 2 H, PhCH₂O), 4.63–4.54 (m, 4 H, PhCH₂O), 4.27–4.23 (m, 1 H, H-2E), 4.19 (dd, 1 H, $J_{2C,3C} = 2.4$ Hz, $J_{3E,4E} = 5.6$ Hz, H-3C), 4.17–4.16 (m, 1 H, H-2C), 4.14–4.11 (m, 1 H, H-4C), 4.09 (dd, 1 H, $J_{2E,3E} = 7.2$ Hz, $J_{3E,4E} = 4.7$ Hz, H-3E), 3.98 (ddd, 1 H, $J_{3E,4E} = 4.7$ Hz, $J_{4E,5Ea} = 2.9$ Hz, $J_{4E,5Eb} = 6.5$ Hz, H-4E), 3.81 (dd, 1 H, $J_{4C,5Ca} = 2.9$ Hz, $J_{5Ca,5Cb} = 12.2$ Hz, H-5Ca), 3.67 (dd, 1 H, $J_{4E,5Ea} = 2.9$ Hz, $J_{5Ea,5Eb} = 12.3$ Hz, H-5Ea), 3.58–3.53 (m, 2 H, H-5Eb, H-5Cb), 3.38 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 138.0 (Ar), 137.8 (Ar), 137.5 (Ar), 128.6 (Ar × 2), 128.4 (Ar × 4), 128.0(9) (Ar), 128.0(5) (Ar × 2), 127.9(1) (Ar), 127.8(4) (Ar × 2), 127.8(2) (Ar), 127.7 (Ar × 2), 107.4 (C-1C), 100.8 (C-1C), 87.0 (C-2C), 84.2 (C-3E), 83.1 (C-4C), 82.5 (C-3C), 82.0 (C-4E), 80.4 (C-2E), 72.8 (PhCH₂O × 2), 72.4 (PhCH₂O), 63.2 (C-5C), 62.1 (C-5E), 54.9 (OCH₃); ESIMS m/z calcd for [C₃₂H₃₈O₉Na]⁺ 589.2408, found 589.2411.

Methyl 2,3-di-O-benzoyl-5-O-*tert*-butyldiphenylsilyl- α -D-arabinofuranosyl-(1 \rightarrow 3)-[2,3-di-O-benzoyl-5-O-*tert*-butyldiphenylsilyl- α -D-arabinofuranosyl-(1 \rightarrow 5)]-2-O-benzyl- α -D-arabinofuranoside (33): The donor **31** (1.80 g, 2.57 mmol) and acceptor **32** (272 mg, 1.07 mmol) were dried under vacuum in the presence of P₂O₅ for 2 h. To this mixture were added CH₂Cl₂ (15 mL) and 4 Å molecular sieves (2.0 g). The solution was allowed to stir at room temperature for 30 min and cooled to –60 °C, followed by the addition of *N*-iodosuccinimide (627 mg, 2.79 mmol) and silver triflate (110 mg, 0.43 mmol). After the solution turned into red/dark brown, triethylamine was added. The reaction mixture was then diluted with CH₂Cl₂ and filtered through Celite. The filtrate was concentrated, and the resulting syrup was purified by chromatography (4:1 hexane–EtOAc) to give **33** (1.32 g, 88%) as a colorless syrup; R_f 0.44 (3:1 hexane–EtOAc); $[\alpha]_D +12.9$ (c 2.0, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 8.06–8.01 (m, 4 H,

Ar), 7.95–7.92 (m, 4 H, Ar), 7.70–7.65 (m, 8 H, Ar), 7.58–7.53 (m, 2 H, Ar), 7.51–7.48 (m, 24 H, Ar), 7.22–7.18 (m, 2 H, Ar), 7.16–7.14 (m, 1 H, Ar), 5.65–5.64 (m, 1 H, H-3D), 5.63–5.61 (m, 1 H, H-3C), 5.53 (d, 1 H, $J_{1D,2D} = 1.4$ Hz, H-2D), 5.43 (d, 1 H, $J_{1C,2C} = 1.7$ Hz, H-2C), 5.27 (s, 2 H, H-1C, H-1D), 4.94 (d, 1 H, $J_{1B,2B} = 1.6$ Hz, H-1B), 4.62–4.57 (m, 2 H, PhCH₂O), 4.44 (ddd, 1 H, $J_{3D,4D} = J_{4D,5Da} = 4.4$ Hz, $J_{5Da,5Db} = 4.6$ Hz, H-4D), 4.36 (ddd, 1 H, $J_{3C,4C} = 3.8$ Hz, $J_{4C,5Ca} = J_{5Ca,5Cb} = 4.7$ Hz, H-4C), 4.27–4.23 (m, 2 H, H-4B, H-3B), 4.10–4.08 (m, 1 H, H-2B), 4.03–3.90 (m, 5 H, H-5Ba, H-5Ca, H-5Cb, H-5Da, H-5Db), 3.85–3.82 (m, 1 H, H-5Bb), 3.35 (s, 3 H, OCH₃), 1.04–0.98 (s, 18 H, *t*-butyl CH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 165.6 (C=O), 165.5 (C=O), 165.2(2) (C=O), 165.2(0) (C=O), 137.5 (Ar), 135.7 (Ar), 135.6(3) (Ar × 2), 135.6(2) (Ar × 2), 135.6(0) (Ar × 2), 135.5 (Ar × 2), 133.4 (Ar), 133.3(3) (Ar), 133.3(1) (Ar), 133.2(8) (Ar), 133.2(4) (Ar), 133.1(9) (Ar), 133.1(6) (Ar), 133.1(4) (Ar), 133.1(1) (Ar), 129.9(5) (Ar × 2), 129.9(3) (Ar × 2), 129.9(1) (Ar × 2), 129.8(6) (Ar × 2), 129.7 (Ar × 2), 129.6(1) (Ar × 2), 129.5(7) (Ar), 129.4(5) (Ar), 129.3 (Ar), 129.2 (Ar), 128.4(8) (Ar × 2), 128.4(6) (Ar × 2), 128.3(9) (Ar × 2), 128.3(6) (Ar × 2), 128.3(2) (Ar × 2), 128.2(8) (Ar × 2), 127.9 (Ar × 2), 127.7 (Ar × 2), 127.6 (Ar × 3), 107.2 (C-1B), 105.8 (C-1D), 105.4 (C-1C), 88.3 (C-2B), 83.5 (C-4C), 83.2 (C-4D), 82.3 (C-2D, C-2C), 81.2 (C-3B), 80.0 (C-4B), 77.3 (C-3D, C-3C), 72.0 (PhCH₂O), 66.6 (C-5B), 63.4 (C-5C), 63.3 (C-5D), 54.9 (OCH₃), 26.8 (*t*-butyl CH₃), 19.3(0) (*t*-butyl C), 19.2(6) (*t*-butyl C); ESIMS *m/z* calcd for [C₈₃H₈₆O₁₇Si₂Na]⁺ 1433.5296, found 1433.5298.

Methyl 2,3-di-*O*-benzyl- α -*D*-arabinofuranosyl-(1→3)-[2,3-di-*O*-benzyl- α -*D*-arabinofuranosyl-(1→5)]-2-*O*-benzyl- α -*D*-arabinofuranoside (34). To a solution of compound 33 (1.32 g, 0.94 mmol) in 1:1 CH₃OH–CH₂Cl₂ (8 mL) was added a 1 M solution of NaOCH₃ in CH₃OH until a pH of 11 was obtained. The reaction was stirred at room temperature for 2 h, neutralized by the addition of AcOH, and concentrated. The residue was dissolved in DMF (20 mL) and THF (3 mL), and then the solution was cooled to 0 °C. NaH 60% in mineral oil (203 mg, 5.08 mmol) and BnBr (0.52 mL, 4.40 mmol) were subsequently added to the solution. The reaction was allowed to stir for 2 h and then quenched by the addition of CH₃OH. The solution was diluted with EtOAc and washed with water. The organic layer was dried (MgSO₄), filtered, and concentrated. The residue was dissolved in THF (10 mL), and then TBAF (2.00 mL, 1 M in THF) was added. The reaction was stirred overnight and concentrated. Chromatography of the residue (1:1 hexane–EtOAc) gave 3 (613 mg, 74% over three steps) as a colorless syrup; *R_f* 0.29 (1:1 hexane–EtOAc); [α]_D +84.8 (c 0.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.40–7.20 (m, 25 H, Ar), 5.17 (s, 1 H, H-1D), 5.13 (d, 1 H, $J_{1C,2C} = 1.5$ Hz, H-1C), 4.96 (d, 1 H, $J_{1B,2B} = 1.5$ Hz, H-1B), 4.62–4.47 (m, 9 H, PhCH₂O), 4.39 (d, 1 H, $J = 11.8$ Hz, PhCH₂O), 4.32 (dd, 1 H, $J_{2B,3B} = 3.8$ Hz, $J_{3B,4B} = 7.4$ Hz, H-3B), 4.26 (ddd, 1 H, $J_{3D,4D} = 2.9$ Hz, $J_{4D,5Da} = 5.2$ Hz, $J_{4D,5Db} = 6.5$ Hz, H-4D), 4.16–4.08 (m, 3 H, H-2D, H-4B, H-4C), 4.04 (dd, 1 H, $J_{1C,2C} = 1.5$ Hz, $J_{2C,3C} = 3.8$ Hz, H-2C), 4.02 (dd, 1 H, $J_{1B,2B} = 1.5$ Hz, $J_{2B,3B} = 3.8$ Hz, H-2B), 3.96 (dd, 1 H, $J_{4B,5Ba} = 3.9$ Hz, $J_{5Ba,5Bb} = 11.6$ Hz, H-5Ba), 3.92 (dd, 1 H, $J_{2D,3D} = 6.6$ Hz, $J_{3D,4D} = 2.9$ Hz, H-3D), 3.86–3.78 (m, 3 H, H-3C, H-5Bb, H-5Db), 3.76 (dd, 1 H, $J_{4C,5Ca} = 2.6$ Hz, $J_{5Ca,5Cb} = 12.2$ Hz, H-5Ca), 3.66 (dd, 1 H, $J_{4D,5Da} = 5.2$ Hz, $J_{5Da,5Db} = 12.1$ Hz, H-5Da), 3.60 (dd, 1 H, $J_{4C,5Cb} = 5.9$ Hz, $J_{5Ca,5Cb} = 12.2$ Hz, H-5Cb), 3.40 (s, 3 H, OCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 133.8 (Ar), 137.7 (Ar), 137.4(7) (Ar), 137.4(6) (Ar), 137.3 (Ar), 128.4(7) (Ar × 2), 128.4(5) (Ar × 3), 128.4(4) (Ar × 3), 128.4(2) (Ar × 2), 128.4(0) (Ar × 2), 127.9(6) (Ar), 127.9(5) (Ar × 2), 127.9(3) (Ar × 2), 127.8(8) (Ar), 127.8(5) (Ar × 3), 127.7(5) (Ar), 127.7(3) (Ar × 3), 106.9 (C-1B), 106.2 (C-1C), 105.9 (C-1D), 88.8 (C-2B), 88.2 (C-2C), 87.6 (C-2D), 83.0(3) (C-3D), 83.0(0) (C-3C), 82.4 (C-4D), 82.0 (C-4C), 81.0 (C-3B), 79.8 (C-4B), 72.3 (PhCH₂O), 72.2 (PhCH₂O), 72.0 (PhCH₂O), 71.9(7)

(PhCH₂O), 71.9(4) (PhCH₂O), 65.0 (C-5B), 62.8 (C-5D), 62.7 (C-5C), 55.0 (OCH₃); ESIMS *m/z* calcd for [C₅₁H₅₈O₁₃Na]⁺ 901.3770, found 901.3764.

Methyl 2,3-di-*O*-benzyl-5-*O*-butyryl- β -*D*-arabinofuranosyl-(1→2)-3-*O*-benzyl-5-*O*-butyryl- α -*D*-arabinofuranoside (35): Isolated as a colorless oil, 96%; *R_f* 0.68 (1:1 hexane–EtOAc); [α]_D –3.7 (c 0.2, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 7.38–7.26 (m, 15 H, Ar), 5.07 (d, 1 H, $J_{1E,2E} = 3.5$ Hz, H-1E), 4.86 (s, 1 H, H-1C), 4.75–4.70 (m, 2 H, PhCH₂O), 4.64–4.50 (m, 4 H, PhCH₂O), 4.30–4.20 (m, 4 H, H-2E, H-5Ea, H-5Ca, H-4E), 4.18–4.13 (m, 3 H, H-5Eb, H-5Cb, H-3C), 4.11–4.08 (m, 2 H, H-2C, H-4C), 3.96 (dd, 1 H, $J_{2E,3E} = 2.7$ Hz, $J_{3E,4E} = 6.1$ Hz, H-3E), 3.40 (s, 3 H, OCH₃), 2.32–2.26 (m, 2 H, acyl CH₂), 2.24–2.20 (m, 2 H, acyl CH₂), 1.68–1.56 (m, 4 H, acyl CH₂ × 2), 0.96–0.87 (m, 6 H, acyl CH₃ × 2); ¹³C NMR (125 MHz, CDCl₃, δ_C) 173.3 (C=O), 173.2 (C=O), 137.8 (Ar), 137.7 (Ar), 137.5 (Ar), 128.5(4) (Ar × 2), 128.4(5) (Ar × 2), 128.3(9) (Ar × 2), 128.1 (Ar), 128.0 (Ar × 2), 127.9 (Ar), 127.8(0) (Ar), 127.7(7) (Ar × 2), 127.7(1) (Ar × 2), 107.0 (C-1C), 100.7 (C-1E), 86.1 (C-2E), 84.1 (C-2C), 83.8 (C-3E), 82.4 (C-4C), 80.0 (C-4E), 78.9 (C-3C), 72.7 (PhCH₂O), 72.5 (PhCH₂O), 72.4 (PhCH₂O), 65.8 (C-5C), 63.7 (C-5E), 54.9 (OCH₃), 35.9(3) (acyl CH₂), 35.8(9) (acyl CH₂), 18.3(4) (acyl CH₂), 18.3(0) (acyl CH₂), 13.6(4) (acyl CH₃), 13.6(0) (acyl CH₃); ESIMS *m/z* calcd for [C₄₀H₅₀O₁₁Na]⁺ 729.3245, found 729.3247.

Methyl 2,3-di-*O*-benzyl-5-*O*-decanoyl- β -*D*-arabinofuranosyl-(1→2)-3-*O*-benzyl-5-*O*-decanoyl- α -*D*-arabinofuranoside (36): Isolated as a colorless oil, 90%; *R_f* 0.48 (2:1 hexane–EtOAc); [α]_D –3.2 (c 0.4, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 7.38–7.26 (m, 15 H, Ar), 5.07 (d, 1 H, $J_{1E,2E} = 3.7$ Hz, H-1E), 4.87 (s, 1 H, H-1C), 4.74–4.70 (m, 2 H, PhCH₂O), 4.64–4.56 (m, 3 H, PhCH₂O), 4.52 (d, 1 H, $J = 11.8$ Hz, PhCH₂O), 4.29 (dd, 1 H, $J_{1C,2C} = 1.2$ Hz, $J_{2C,3C} = 2.9$ Hz, H-2C), 4.27 (dd, 1 H, $J_{4E,5Ea} = 3.7$ Hz, $J_{5Ea,5Eb} = 3.7$ Hz, H-5Ea), 4.25–4.20 (m, 2 H, H-5Ca, H-4E), 4.17–4.12 (m, 3 H, H-5Eb, H-5Cb, H-3C), 4.11–4.08 (m, 2 H, H-2E, H-4C), 3.95 (dd, 1 H, $J_{2E,3E} = 2.9$ Hz, $J_{3E,4E} = 6.2$ Hz, H-3E), 3.38 (s, 3 H, OCH₃), 2.34–2.27 (m, 2 H, acyl CH₂), 2.26–2.18 (m, 2 H, acyl CH₂), 1.64–1.50 (m, 4 H, acyl CH₂ × 2), 1.34–1.20 (m, 24 H, acyl CH₂ × 12), 0.92–0.86 (m, 6 H, acyl CH₃ × 2); ¹³C NMR (125 MHz, CDCl₃, δ_C) 173.5 (C=O), 173.3 (C=O), 137.8 (Ar), 137.7 (Ar), 137.5 (Ar), 128.5(4) (Ar × 2), 128.4(5) (Ar × 2), 128.3(8) (Ar × 2), 128.1 (Ar), 128.0 (Ar × 2), 127.9 (Ar), 127.7(9) (Ar), 127.7(7) (Ar × 2), 127.6(9) (Ar × 2), 107.0 (C-1C), 100.7 (C-1E), 86.1 (C-2E), 84.1 (C-2C), 83.8 (C-3E), 82.5 (C-4C), 80.0 (C-4E), 78.9 (C-3C), 72.7 (PhCH₂O), 72.5 (PhCH₂O), 72.4 (PhCH₂O), 65.9 (C-5C), 63.8 (C-5E), 54.9 (OCH₃), 34.1 (acyl CH₂), 31.8(7) (acyl CH₂), 31.8(5) (acyl CH₂), 29.4(5) (acyl CH₂), 29.4(1) (acyl CH₂), 29.3(0) (acyl CH₂), 29.2(8) (acyl CH₂), 29.2(5) (acyl CH₂), 29.1(5) (acyl CH₂), 29.1(4) (acyl CH₂), 24.8(7) (acyl CH₂), 24.8(3) (acyl CH₂), 22.7 (OCH₃), 14.1 (acyl CH₃ × 2); ESIMS *m/z* calcd for [C₅₂H₇₄O₁₁Na]⁺ 897.5127, found 897.5122.

Methyl 2,3-di-*O*-benzyl-5-*O*-palmitoyl- β -*D*-arabinofuranosyl-(1→2)-3-*O*-benzyl-5-*O*-palmitoyl- α -*D*-arabinofuranoside (37): Isolated as a colorless oil, 82%; *R_f* 0.56 (4:1 hexane–EtOAc); [α]_D –4.3 (c 0.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 7.40–7.22 (m, 15 H, Ar), 5.06 (d, 1 H, $J_{1E,2E} = 3.6$ Hz, H-1E), 4.86 (s, 1 H, H-1C), 4.74–4.69 (m, 2 H, PhCH₂O), 4.64–4.55 (m, 3 H, PhCH₂O), 4.52 (d, 1 H, $J = 11.7$ Hz, PhCH₂O), 4.29–4.27 (m, 2 H, H-2C, H-5Ca), 4.26–4.19 (m, 2 H, H-5Ea, H-4C), 4.16–4.07 (m, 5 H, H-5Eb, H-5Cb, H-3E, H-2E, H-4E), 3.94 (dd, 1 H, $J_{2C,3C} = 2.4$ Hz, $J_{3C,4C} = 6.0$ Hz, H-3C), 3.37 (s, 3 H, OCH₃), 2.32–2.27 (m, 2 H, acyl CH₂), 2.24–2.20 (m, 2 H, acyl CH₂), 1.62–1.50 (m, 4 H, acyl CH₂ × 2), 1.34–1.20 (m, 48 H, acyl CH₂ × 24), 0.92–0.86 (m, 6 H, acyl CH₃ × 2); ¹³C NMR (125 MHz, CDCl₃, δ_C) 173.5 (C=O), 173.3 (C=O), 137.8 (Ar), 137.7 (Ar), 137.5 (Ar), 128.5(3) (Ar × 2), 128.4(5) (Ar × 2), 128.3(8) (Ar × 2), 128.1 (Ar), 128.0 (Ar × 2), 127.8(5) (Ar), 127.7(9) (Ar),

(d, 1 H, $J_{1C,2C} = 0.8$ Hz, H-1C), 5.14 (d, 1 H, $J_{1D,2D} = 0.9$ Hz, H-1D), 4.92 (d, 1 H, $J_{1B,2B} = 0.9$ Hz, H-1B), 4.59–4.42 (m, 9 H, PhCH₂O), 4.37 (d, 1 H, $J = 11.8$ Hz, PhCH₂O), 4.29–4.22 (m, 4 H, H-3B, H-5Ca, H-5Da, H-4C), 4.21–4.16 (m, 3 H, H-5Cb, H-5Db, H-4D), 4.12 (ddd, 1 H, $J_{3B,4B} = 7.0$ Hz, $J_{4B,5Ba} = 4.7$ Hz, $J_{4B,5Bb} = 2.6$ Hz, H-4B), 4.09 (dd, 1 H, $J_{1C,2C} = 1.1$ Hz, $J_{2C,3C} = 3.4$ Hz, H-2C), 4.02 (dd, 1 H, $J_{1D,2D} = 1.1$ Hz, $J_{2D,3D} = 3.4$ Hz, H-2D), 3.97 (dd, 1 H, $J_{1B,2B} = 1.1$ Hz, $J_{2B,3B} = 3.3$ Hz, H-2B), 3.94 (dd, 1 H, $J_{4B,5Ba} = 4.7$ Hz, $J_{5Ba,5Bb} = 11.8$ Hz, H-5Ba), 3.87–3.83 (m, 2 H, H-3C, H-3D), 3.77 (dd, 1 H, $J_{4B,5Bb} = 2.6$ Hz, $J_{5Ba,5Bb} = 11.8$ Hz, H-5Bb), 3.37 (s, 3 H, OCH₃), 2.30–2.22 (m, 4 H, acyl CH₂ × 2), 1.60–1.50 (m, 4 H, acyl CH₂ × 2), 1.36–1.20 (m, 72 H, acyl CH₂ × 36), 0.92–0.84 (m, 6 H, acyl CH₃ × 2); ¹³C NMR (125 MHz, CDCl₃, δ_C) 173.6 (C=O), 173.5 (C=O), 137.8 (Ar), 137.6(4) (Ar), 137.5(7) (Ar), 137.4 (Ar), 137.3 (Ar), 128.5 (Ar × 2), 128.4(3) (Ar × 2), 128.3(9) (Ar × 3), 128.3(5) (Ar × 2), 127.9(6) (Ar), 127.9(2) (Ar × 3), 127.8(9) (Ar × 3), 127.8(6) (Ar × 3), 127.7(9) (Ar), 127.7(6) (Ar × 2), 127.7(3) (Ar × 2), 127.7(1) (Ar), 107.1 (C-1B), 106.5 (C-1C), 105.5 (C-1D), 88.3 (C-2B), 88.1(3) (C-2C), 88.0(5) (C-2D), 83.4(3) (C-3C), 83.3(6) (C-3D), 80.8 (C-4B), 80.5 (C-3B), 79.2 (C-4D), 79.0 (C-4C), 72.2 (PhCH₂O), 72.1 (PhCH₂O), 72.0 (PhCH₂O × 2),

71.8 (PhCH₂O), 65.8 (C-5B), 63.3 (C-5D), 63.2 (C-5C), 54.8 (OCH₃), 34.1 (acyl CH₂), 34.0 (acyl CH₂), 31.9 (acyl CH₂), 29.7(2) (acyl CH₂), 29.6(7) (acyl CH₂), 29.5 (acyl CH₂), 29.4 (acyl CH₂), 29.3(4) (acyl CH₂), 29.3(2) (acyl CH₂), 29.2 (acyl CH₂), 24.8 (acyl CH₂), 22.7 (acyl CH₂), 14.1 (acyl CH₃ × 2); ESIMS *m/z* calcd for [C₉₅H₁₄₂O₁₅Na]⁺ 1546.0241, found 1546.0244.

Acknowledgment. This work was supported by the Alberta Ingenuity Centre for Carbohydrate Science and the Natural Sciences and Engineering Research Council of Canada. M.R.R. is supported by an Alberta Heritage Foundation for Medical Research Ph.D. studentship.

Supporting Information Available: Examples of simulated ¹H NMR spectra used to determine the coupling constants in Tables 3 and 4; computational methods and Cartesian coordinates for methyl 5-*O*-acetyl- α -D-arabinofuranoside used to determine eqs 6 and 7; ¹H and ¹³C NMR spectra of new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.